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### Synthesis of Jasplakinolide Analogues Containing a Novel w-Amino Acid

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Dedicated to Prof. R. R. Schmidt for his 70th birthday

Abstract: The synthesis of the  $\omega$ -amino acid 4 is described utilizing a two-dimensional synthesis strategy combined with an enzymatic differentiation of homotopic ester groups. The amino acid 4 features two non-bonded interactions that result in conformational constraints on a cyclic construct. This amino acid was incorporated into the four macrolactams 17, 22, 31, and 37. The ring in 17 and 22 is 18-membered, whereas 31 and 37 have a 19-mem-

### Introduction

Constraining the conformation of a peptide fragment by incorporating it in a macrocyclic structure represents an important strategy for enhancing both binding strength and selectivity. In addition, this maneuver can suppress unwanted proteolysis. Studying the solution conformation of such a macrocyclic construct can provide important information on the peptide surface structure and the area that is presented to a receptor. The application of this strategy is common in medicinal chemistry.[1] In a class of natural products, such as the cyclodepsipeptides, the presence of the macrocyclic ring most likely also improves binding and stability. Besides a macrocyclic ring, cyclodepsipeptides are characterized by the presence of at least one ester bond because they contain

bered ring. The pairs with the same ring size differ in a N-methyl group. For the larger macrolactams (31 and 37) conformational analysis showed that the macrocyclic rings are somewhat more rigid than in the natural lead, the depsipeptide jasplakinolide.

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Nevertheless, their conformations are comparable to the natural product. There are no intramolecular hydrogen bonds, neither is the cis-rotamer populated in the N-methyl compound 37. Due to the increased flexibility of the smaller macrolactams 17 and 22 and signal overlap, a distinct solution structure could not be obtained for these compounds. The amino acid 4 should be useful for restricting the conformation of other small peptides.

a hydroxy acid. Furthermore, they contain unusual amino acids, which may be extended, N-methylated, hydroxylated, or halogenated. Occasionally, they contain fragments from other biosynthetic pathways, for example polyketides. The substituents on the polyketide fragment might be used as conformational control elements. An illustrative example is the depsipeptide jasplakinolide A (1) (Figure 1). This natural product, which was isolated from the sponge Jaspis  $sp.$ ,<sup>[2,3]</sup> shows potent antifungal, insecticidal, and antitumor activity. Related to the latter activity is its use as a tool in cytoskeletal research, since it stabilizes F-actin.[4] A related compound is the cyclodepispeptide geodiamolide  $A$ , [5,6] (2), which has somewhat different activity.<sup>[7]</sup>

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Besides the tripeptide fragment, the  $\omega$ -hydroxyacid 3 is part of these depsipeptides (Figure 2). This hydroxy acid contains four methyl groups in a 1,3-distance giving rise to two syn-pentane interactions and one 1,3-allylic interaction.[8] As a result of this the carboxyl and hydroxyl groups at the end of the chain point in one direction, thereby allowing bridging with a peptide fragment. While the synthesis of the hydroxy acid 3 is feasible,<sup>[9]</sup> the preparation of a larger amount is quite costly. We therefore turned to the design of simpler analogues of this hydroxy acid that also contains conformation controlling elements.

The design of the amino acid 4 followed from looking at the conformational control elements in the hydroxy acid 3 (Figure 2). Thus, the 1,3-allylic interaction around the central trisubstituted double bond should position the allylic C H in an eclipsed orientation to the double bond.<sup>[8]</sup> As a consequence, the methyl group at C-6 will point downwards and orient the 2-hydroxypropyl terminus to the other side, out of the plane of the double bond. The conformational situation at the carboxyl terminus seems to be less defined. Nevertheless, avoidance of syn-pentane interactions between C2–Me and C4–Me will cause the carboxyl group to point out of the plane of the central double bond as well. Our design plan then called for a rigidification of the vinylic C5 C6 bond. Accordingly, the allylic H was replaced with a two-carbon segment (see dashed lines in Figure 2), resulting in a meta-disubstituted aryl core. This simplification removes the stereocenter at C-6.



Figure 2. Design of the novel  $\omega$ -amino acid 4.

In order to probe the design process, conformational search runs (Macromodel 7.0, MM2\* force field, 1000 starting structures) were carried out on hydroxy acid 3 and the amino acid 4 (without the N-Boc protecting group). The search for the hydroxy acid 3 found four conformers within 4.184 kJ mol<sup>-1</sup>  $(1.0 \text{ kcal mol}^{-1})$  of the minimum  $(E=$  $36.14 \text{ kJ} \text{mol}^{-1}$ ). The lowest conformer **3a** has the allylic

 $C-H$  eclipsing the  $C4-Me$ , but only the carboxyl group is pointing out of the plane of the double bond (Figure 3). In the next lowest conformer **3b** ( $\Delta E = 0.93$  kJ mol<sup>-1</sup>) 6-H, surprisingly, is *anti* to the C4–Me. Basically, C4–Me is bisecting



Figure 3. Calculated low energy conformations of the hydroxy acid 3 and the amino acid 4 (Chem3D representations).

the angle C6-Me/C-6/C-7. This orients both termini to the other side of the central  $\pi$ -system. Conformers 3c ( $\Delta E=$ 2.00 kJ mol<sup>-1</sup>) and **3d**  $(\Delta E = 4.10 \text{ kJ} \text{ mol}^{-1})$  actually match the expected conformation. Thus, 6-H eclipses the vinylic methyl and the termini extend nicely to one side. For the amino acid 4 we found two conformers within  $4.184 \text{ kJ} \text{mol}^{-1}$ of the absolute minimum  $(E=35.96 \text{ kJ} \text{ mol}^{-1})$  In all cases, the termini are oriented more or less orthogonal to the plane of the aryl ring. In the second lowest conformer 4b  $(\Delta E = 2.45 \text{ kJ} \text{ mol}^{-1})$  the termini point to the opposite side of the aryl ring. Conformer **4b** does have a striking similarity to conformer  $3c$  of the hydroxy acid. Most likely electrostatic interaction and hydrogen bonding cause a substantial gain in energy if both groups point to the same side. By looking at several of the calculated minima, it seems that the conformation of the aryl analogue is more ordered and less flexible.

An overlay of conformers 3c and 4b shows a decent overlap validating our original design (Figure 4). In another run, the configuration at the amino bearing carbon was inverted. In this case we found also several reasonable conformers, in which both termini extend from the same face of the aryl ring.

It should be mentioned that alternative strategies for constraining the conformation of peptides are known. Frequently, this involves the use of designed amino acids containing a

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Figure 4. Overlay of the calculated conformers 3c and 4b (grey, hydroxy acid; black, aromatic amino acid).

cyclic backbone.[10] In contrast to our work, a recent paper describes the synthesis of three jasplakinolide analogues in which the  $\omega$ -hydroxy acid was replaced with  $\omega$ -amino acids, for example 6-amino hexanoic acid, lacking conformational constraints.[11]

The synthesis of the amino acid 4 began with the commercially available 1,3-bis(bromomethyl)benzene (6), which was subjected to a double alkylation<sup>[12,13]</sup> with the propionyloxazolidinone<sup>[14]</sup> **5** (Scheme 1). This way the C-2 symmetric alkylation product 7 was obtained in satisfactory yield (58%). Subsequently, the chiral auxiliary was removed by hydrolysis



Scheme 1. Preparation of the N-protected  $\omega$ -amino acid 4. a) LDA (2.4 equiv), THF,  $-78^{\circ}$ C, then add 6 (58%); b) LiOH,  $H_2O_2$ , 23 $^{\circ}$ C, 5 h  $(95\%)$ ; c) MeOH, DCC, DMAP, CH<sub>2</sub>Cl<sub>2</sub>, 23 °C (71%); d) PLE, H<sub>2</sub>O, pH 7.5, 12 h, 23 °C (64%); e) (PhO)<sub>2</sub>P(O)N<sub>3</sub>, toluene, 23 °C, 1 h, then reflux, 3 h, add tBuOH, reflux, 20 h (72%); f) NaOH, THF/H<sub>2</sub>O, 23°C, 12 h  $(80\%)$ ; PLE = pig liver esterase.

to the diacid 8, which was in turn converted to the dimethyl ester 9 via DCC-mediated esterification. This maneuver was necessary in order to allow for a monohydrolysis of the homotopic ester groups. This could be achieved by esterase-induced hydrolysis.<sup>[15]</sup> Treatment of the monoacid 10 with diphenyl-diphosphoryl azide followed by heating of the reaction mixture in the presence of tert-butanol affected a Curtius rearrangement resulting in the N-Boc protected wamino acid ester 11. Basic hydrolysis of the ester group led

to the desired acid 4. This route secures this novel amino acid in gram quantities.

With the acid 4 in hand, the synthesis of various macrocycles was targeted, in which the two ends of acid 4 are bridged with some tripeptide fragments. Our goal was to prepare pairs of tripeptide fragments with one of them carrying a Nmethyl group at the middle amino acid. The conformational analysis of the corresponding macrocycles should provide some hints on the mutual influence of the parts contained in the macrocycle. Thus, tripeptide 15 was assembled by a classical Boc strategy (Scheme 2). That is, DCC-mediated condensation of the phenylalanine derivative 12 with N-Boc-D-



Scheme 2. Preparation of the geodiamolide analogue 17. a) DCC, HOBt, amino acid 13, THF,  $0 \to 23^{\circ}C$ , 12 h (80%); b) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 23 °C; c) EDCI, HOBt, Boc-l-Ala-OH, Et<sub>3</sub>N, THF/CH<sub>2</sub>Cl<sub>2</sub> 5:1,  $0 \rightarrow 23^{\circ}$ C, 16 h (75%); d) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 23°C; e) TBTU, HOBt, amino acid 4, iPr<sub>2</sub>NEt, DMF, 23 °C, 3 h (95%, crude); f) NaOH, THF/H<sub>2</sub>O, 23 °C, 3 h; g) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 23<sup>°</sup>C, 1 h; h) TBTU, HOBt,  $iPr_2NEt$ , DMF, 23<sup>°</sup>C, 14 h (50%, three steps).

alanine 13 gave the dipeptide 14. After cleavage of the Boc protecting group, coupling of the free amine with N-Boc-lalanine with the coupling reagent 1-ethyl-3-(3-dimethylaminopropyl) carbodiimide hydrochloride (EDCI) provided compound 15 in 75% yield. Boc cleavage (TFA) and condensation of the resulting amine with the amino acid 4 led to the acyclic tetrapeptide 16. Hydrolysis of the methyl ester, removal of the Boc group and macrolactam formation with 2-(1H-benzotriazole-1-yl))-1,1,3,3-tetramethyluronium tetrafluoroborate (TBTU) in DMF  $(0.001 \text{ m})$  gave rise to the jasplakinolide (geodiamolide) analogue 17 in 50% yield.

In a similar manner, the tripeptide 20 was assembled (Scheme 3). Here, N-Boc-N-methyl-D-alanine $[16, 17]$  became the central amino acid of the tripeptide fragment. For the formation of the peptide bond to the N-methylated amine, the coupling reagent bromotrispyrrolidinophosphonium hexafluorophosphate (PyBroP) came to use. After liberation of



Scheme 3. Preparation of the geodiamolide analogue 22. a) DCC, HOBt, amino acid 18, THF,  $0 \to 23^{\circ}C$ , 12 h (70%); b) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 23<sup>°</sup>C; c) PyBroP,  $iPr_2NEt$ , N-Boc-L-Ala-OH (13), CH<sub>2</sub>Cl<sub>2</sub>, 23<sup>°</sup>C, 1 h (61%); d) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 23<sup>°</sup>C; e) EDCI, HOBt, amino acid 4, CH<sub>2</sub>Cl<sub>2</sub>/THF, 0  $\rightarrow$ 23°C, 7 h (52%); f) NaOH, THF/H<sub>2</sub>O, 23°C, 3 h (90%); g) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 23<sup>°</sup>C, 1 h; h) TBTU, HOBt,  $iPr_2NEt$ , DMF, 23<sup>°</sup>C, 14 h (65%).

the terminal amine, EDCI-mediated condensation with the amino acid 4 provided the seco-compound 21. Ester hydrolysis, cleavage of the Boc protecting group and macrolactam formation delivered the geodiamolide analogue 22 with a Nmethyl amide group.

The difference between jasplakinolide and geodiamolide is the presence of a  $\beta$ -amino acid in the former one. While several methods for the synthesis of  $\beta$ -amino acids exist, the methyl ether derivative 27 was prepared via an asymmetric alkylation reaction of the oxazolidinone<sup>[18]</sup> 23 by using tertbutyl-bromoacetate as electrophile (Scheme 4).<sup>[19]</sup> The alkylation product 24 was hydrolyzed to the acid 25, which in



Scheme 4. Synthesis of the  $\beta$ -amino acid 27 via asymmetric alkylation and Curtius rearrangement. a)  $\text{NaN}(\text{SiMe}_3)$ , THF,  $-78 \text{°C}$ , 2.5 h, then add BrCH<sub>2</sub>CO<sub>2</sub>tBu, -78°C, 3 h (71%); b) H<sub>2</sub>O<sub>2</sub>, LiOH, THF, 0°C, 5 h  $(78\%)$ ; c)  $(PhO)_{2}P(O)N_{3}$ , Et<sub>3</sub>N, toluene, reflux, add fluorenyl-OH, reflux, 3 h (45%); d) Et<sub>2</sub>NH, THF,  $0 \rightarrow 23^{\circ}$ C, 12 h (60%).

turn was subjected to a Curtius rearrangement. The resulting isocyanate was reacted with 9H-fluoren-9-ylmethanol to yield the protected b-amino acid 26. Subsequent treatment of the urethane 26 gave free amine 27.

The carboxyl-protected amino acid 27 was then condensed with N-Fmoc-D-tryptophan to yield compound 28 (Scheme 5). Deprotection and another condensation provided the tripeptide 29. Thereafter, liberation of the terminal amine and condensation with the  $\omega$ -amino acid 4 gave the protected seco-compound 30. Both protecting groups could now be removed in one step by using trifluoroacetic acid. After concentration of the reaction mixture, macrolactam formation with TBTU in the presence of HOBt at room temperature led to compound 31 in very good yield.



Scheme 5. Synthesis of the tripeptide 29, coupling with amino acid 4, and formation of macrocycle  $31.$  a) Fmoc-D-Trp-OH, DCC, HOBt, THF,  $-10$  $\rightarrow$  0°C, 12 h (91%); b) Et<sub>2</sub>NH, THF, 0  $\rightarrow$  23°C, 80 min (78%); c) Fmoc-L-Ala-OH, DCC, HOBt, THF, 0°C, 12 h (97%); d) Et<sub>2</sub>NH, THF, 0  $\rightarrow$ 23<sup>°</sup>C, 80 min (85%); e) DCC, HOBt, THF, amino acid 4,  $-20 \rightarrow 23$ <sup>°</sup>C, 8.5 h (92%); f) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 0  $\rightarrow$  23<sup>°</sup>C, 1.5 h; g) TBTU, HOBt,  $iPr<sub>2</sub>NEt$ , DMF, 23 °C, 14 h (92%, two steps).

In order to reach the N-methyl analogue 37 of macrocycle 31, the assembly of the tripeptide part needed to be changed, since N-methylation was not compatible with the Fmoc protecting group. Accordingly, the known tryptophan derivative<sup>[3d]</sup> 32 was deprotected at the nitrogen and elongated with N-Boc-l-alanine (Scheme 6). At this point, the methyl ester at the C terminus was cleaved, followed by amide formation of the resulting acid with the amino acid 27. The selective cleavage of the N-Boc group could be achieved with TBDMSOTf and 2,6-lutidine.<sup>[20]</sup> The remainder of the synthesis proceeded as before. Thus, DCC-mediated condensation of the resulting amine with the amino acid 4 gave the tetrapeptide 36. Deprotection and macrolac-



Scheme 6. Synthesis of the macrocycle 37 with a N-methyl group at the tryptophan amino acid. a) TFA, CH<sub>2</sub>Cl<sub>2</sub>, 23°C, 1 h (68%); b) Boc-L-Ala-OH, DCC, HOBt, THF,  $0 \rightarrow 23^{\circ}$ C, 14 h (59%); c) NaOH, THF, 23 $^{\circ}$ C, 2 h (74%); d) amino acid 27, DCC, HOBt,  $-20 \rightarrow 0$ °C, 12 h (96%); e) TBDMSOTf, 2,6-lutidine, CH<sub>2</sub>Cl<sub>2</sub>, 0°C, 1 h, then add H<sub>2</sub>O (56%); f) amino acid 4, DCC, HOBt, THF,  $-20 \rightarrow 23^{\circ}$ C, 18.5 h (60%); g) TFA,  $CH_2Cl_2$ ,  $0 \rightarrow 23^{\circ}C$ , 1.5 h; h) TBTU, HOBt,  $iPr_2NEt$ , 23 ${}^{\circ}C$ , 14 h (45%, two steps).

tam formation led to 37. However, in this case the yield for the macrocyclization was lower.

### Conformational Studies

Aside from variations in the side-chains, the major difference of the four analogues 17, 22, 31, and 37 is the replacement of the polypropionate subunit with a  $m$ -xylyl containing amino acid 4. The 19-membered systems 31 and 37 are structurally closely related to the natural compound jasplakinolide. Compound 31 possesses a secondary amide in position 16 (Figure 5). <sup>1</sup>H and <sup>13</sup>C resonance assignments via DQFCOSY, ROESY/NOESY and HMQC spectra were performed in  $[D_6]$ DMSO and gave a single signal set for 31 and a doubled signal set for 37 with the trans isomer of the Nmethylated amide populated for  $> 99\%$ .

One large and one small  ${}^{3}J_{\text{HH}}$  coupling constant (Table 1) for the methylene groups in positions 2, 6, and 10 of 31 and in positions 6 and 10 for 37 are typical for a well-defined gauche–anti orientation with preference for the major rotamer(s) as distinguished by characteristic ROE signals. This allowed the assignment of the  $pro-R$  and  $pro-S$  protons of the methylene groups. Due to signal overlay, a statement on the methylene protons in position 2 of 37 was not possible.

With ROESY and NOESY measurements, characteristic proton–proton interactions could be determined and trans-



Figure 5. Structures for the natural compound jasplakinolide and the 19 membered ring analogue 31 with the numbering systems used.

Table 1.  ${}^{3}J_{\text{HH}}$  coupling constants in Hz of the methylene protons in 17, 22, 31 and 37. Due to identical chemical shift for  $H10<sup>h</sup>/H10<sup>t</sup>$  of 17 and  $H2<sup>h</sup>/H2<sup>t</sup>$  of 37, the coupling constants could not be determined.

|                              | 17   | 22               |  | 31   | 37   |
|------------------------------|------|------------------|--|------|------|
| $^{2\!}J_{\rm{H1^h,H1^t}}$   | 13.5 | 14.0             | $^{2}J^{\phantom{+}}_{\mathrm{H2^h,H2^t}}$ | 13.6 | n.d. |
| ${}^3\!J_{\rm H1^h,H2}$      | n.d. | 12.7             | ${}^3\!J_{\text{H2^h,H1}}$                 | 4.3  | n.d. |
| ${}^3J_{\rm H1^1,H2}$        | 3.7  | 3.6              | ${}^3\! J_{\rm H2^t\!,H1}$                 | 10.4 | n.d. |
| ${}^2\!J_{{\rm H6^h,H6^t}}$  | 13.1 | 13.1             | $^2J_{\rm{H6^h,H6^t}}$                     | 13.0 | 12.9 |
| ${}^3\!J_{{\rm H6^h,H5}}$    | 6.1  | $~1$ (from COSY) | $^{3}J_{\rm{H6^h,H5}}$                     | 10.6 | 9.0  |
| $^{3}J_{\rm{H6^{t},H5}}$     | 4.1  | $~5$ (from COSY) | ${}^3\! J_{\rm H6^t\!,H5}$                 | 3.7  | 4.7  |
| $^{2\!}J_{\rm{H10^h,H10^t}}$ | n.d. | 12.6             | $^{2}J_{\mathrm{H10^h,H10^t}}$             | 11.0 | 14.0 |
| ${}^3\! J_{\rm H10^h,H11}$   | n.d. | 11.3 (via H11)   | ${}^3\! J_{\rm H10^h,H11}$                 | 2.7  | 3.1  |
| ${}^3\!J_{\rm H10^t,H11}$    | n.d. | 4.0              | ${}^3\!J_{\rm H10^{\rm t},H11}$            | 13.7 | 10.3 |

ferred into proton–proton distances by integration. They are in very good agreement for compounds 31 and 37 corresponding to a minor influence of the N-methylation in position 16 on the overall structure. No intensive cross signals were found between  $H\alpha$  protons of adjacent amino acids, thus proving trans conformation for all amide bonds. The methylated amide in 37 leads to an energy barrier that gives rise to a separated signal set for the cis-amide. The signal set for the cis isomer is populated to less than 1%. The significant preference for the trans rotamer, to such an extent unusual for a tertiary amide, was found for the natural compound jasplakinolide as well,<sup>[21]</sup> thus emphasizing the good analogy of synthetic analogue and natural product.

The dynamical properties of the whole macrocycles 31 and 37 is consistent including the side-chains having the same dynamical behavior as the macrocyclic ring. The newly inserted m-xylyl unit performs an oscillating movement as ROESY data yield only mean proton–proton distances for several possible orientations of the phenyl ring in relation to the macrocyclic ring.

Via temperature measurements, which gave temperature coefficients for the amidic NH protons in the range of  $-3.7$ to  $-6.0$  ppb  $K^{-1}$ , transannular hydrogen bonds could be excluded, but nevertheless, as coupling constants and ROESY data show, the 19-membered analogues 31 and 37 are rigid macrocyclic structures.

Within a MD simulation, calculated structures of 31 and 37 are very well comparable with each other, the only difference being the orientation of the NH proton respective to

the N-methyl group in position 16 with the proton in 31 pointing into the middle of the macrocyclic ring and the methyl group in 37 oriented towards the lower side of the ring (Figure 6).



Figure 6. Energy-minimized structures for 31 and 37 after a 100 ps MD simulation at 300 K. In green, torsion with the largest difference for 31 and 37 with the N-methyl group in 37 oriented onto the lower side of the ring and the NH proton in position 16 of 31 pointing into the macrocyclic ring.

The results presented above are in accordance with NMR structural investigations of jasplakinolide.<sup>[21]</sup> Jasplakinolide does not possess any hydrogen bonds, nor a higher fraction of *cis*-amide bonds either. The  $3J$  coupling constants yield a more flexible structure for the natural analogue in comparison to the compounds 31 and 37, however, the difference is small. The orientation of the two aromatic side-chains were not determined in detail in the study presented here. But a possible "tweezer" structure as stated in literature<sup>[21]</sup> cannot be supported by NOEs between side-chain protons.

The 18-membered rings of compounds 17 and 22 show several structural differences when compared to jasplakinolide. They are rather analogues of geodiamolide with a difference in the position of the aromatic amino acid (Figure 7). $[5]$ 

 ${}^{1}$ H and  ${}^{13}$ C signal assignment was done by DQFCOSY, ROESY/NOESY, and HMQC spectra in  $[D_6]$ DMSO with results comparable to the 19-membered macrocycles: a

single signal set for 17, the non-methylated amide and a doubled signal set for 22 with the cis isomer of the N-methylated amide populated to less than 1%.

Investigation of the NOESY/ROESY data results in mean values of proton– proton distances being significant for the fast exchange of several conformers. Thus the 18-membered rings 17 and 22 are more flexible than the systems described above and a



Figure 7. Structures for the natural compound geodiamolide and the 18 membered ring analogue 17 with the numbering systems used.

ROESY data again reveal an oscillatory movement for the  $m$ -xylyl unit like in 31 and 37. In case of the geodiamolide analogues 17 and 22 there are no consistent dynamics for the whole molecule, but rather an independent dynamical behavior of the side-chains, that is, benzyl and methyl groups, as seen in ROESY spectra with cross signals of different sign.

The methylene groups in position 1 of 17 and in positions 1 and 10 of 22 exhibit one large and one small  $\beta J_{\text{HH}}$  coupling constant (Table 1) defining a preferred gauche–anti orientation, whereas in position 6 the coupling constants possess mediated values upon the presence of several rotamers.

Temperature measurements yield low temperature coefficients for NH4 and NH13 for both structures  $(NH4: -1.0,$  $-1.1$  ppb K<sup>-1</sup>, respectively; NH13:  $+0.4$ ,  $-1.1$  ppb K<sup>-1</sup>, respectively) with a high probability for those NH protons to take part in an intramolecular hydrogen bond.

When calculating only a partial structure for 22 with NOESY/ROESY data a bII'-turn-like structure is obtained for the macrocyclic part containing the phenylalanine unit. For this case NH4 is part of a transannular hydrogen bond with CO(15) as partner, whereas for NH13 there is no geometrically reasonable arrangement for the formation of a hydrogen bond (Figure 8). At least for this section of the macrocyclic ring the ROESY data obtained are in accordance with the distances typical for  $\beta$ II' turns.



Figure 8. Comparison of the calculated partial structure of  $22$  with a  $\beta$ II'-turn structure.

preferred structure can not be calculated on the basis of experimental ROE data. ROESY data confirm a trans configuration for all amide bonds in 17 and 22. Additionally the

Nevertheless, the 18-membered rings 17 and 22 are more flexible compared with 31, 37, or jasplakinolide and an equilibrium of several fast-exchanging conformers is existent.

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### Biological Studies

The four analogues were tested for cytotoxicity against two cell lines.[22] Compound 37 turned out to be the most active one. It inhibits the growth of L929 mouse fibroblasts with an  $IC_{50}$  of 25  $\mu$ g mL<sup>-1</sup>. For the ovary cancer cell line SKOV-3 the  $IC_{50}$  amounts to 20  $\mu$ gmL<sup>-1</sup>. Compound 31 did show weak activity against the L929 cells. The geodiamolide analogues 17 and 21 were devoid of any activity. While the activity of 37 is moderate it does show that the N-Me group makes a difference. Most remarkably, at a concentration of  $40 \mu g \text{mL}^{-1}$ , the assay with SKOV-3 cells shows many cells with two nuclei. This is a clear indication that the actin polymerization is disturbed. Based on the activity data it can be said that analogue 37 is the most jasplakinolide-like among the four compounds.

### Summary

In this paper we describe the synthesis of a novel  $\omega$ -amino acid 4 that incorporates conformational constraints due to non-bonded interactions (syn-pentane interactions). The design was guided by the polypropionate sector of the depsipeptide jasplakinolide. This novel amino acid was prepared by a double alkylation, enzyme-mediated hydrolysis of homotopic ester groups, and a Curtius rearrangement on the carboxylic acid. The amino acid was incorporated into the four macrolactams 17, 22, 31, and 37. The former two feature an 18-membered macrocycle, whereas the latter two have a 19-membered ring. For the two larger ones, conformational analysis showed that the macrocyclic rings are more rigid than the jasplakinolide ring, but all in all their conformations are very well comparable to the natural product. Like stated for jasplakinolide in the literature,[21] the 19 membered analogues exhibit neither intramolecular hydrogen bonds, nor is the cis-rotamer populated in case of Nmethylation (37). The conformational features of the 18 membered (geodiamolide) analogues are quite different. The macrocyclic ring of compounds 17 and 22 is more flexible than the other investigated systems. Due to the increased flexibility and signal overlap a distinct solution structure for 17 and 22 could not be gained. Whether the ring size or the additional aromatic side-chains in 31 and 37 cause a stabilizing effect on the macrocyclic system could not be determined. But both features are differing for 17 and 22 and might thus be an explanation for the higher flexibility of these smaller macrocycles. In addition, both N-methylated analogues  $(22 \text{ and } 37)$  populate the *trans*-amide conformer to more than 99%.

The amino acid 4 could serve as a novel workbench for restricting the conformation of small peptides. Furthermore, the aryl group might serve as a handle for attachment of derived macrocycles to a solid surface. The incorporation of the m-xylene subunit into the amino acid indicates that small structural modifications can have subtle effects on a macrocyclic structure. Finally, it should be noted that the N-

methyl analogues 22 and 37 only populate the trans-amide conformer.

### Experimental Section

**General methods:** <sup>1</sup>H and <sup>13</sup>C NMR: Bruker Avance 400, spectra were recorded at 295 K either in CDCl<sub>3</sub>,  $C_6D_6$ , or [D<sub>6</sub>]acetone; chemical shifts are calibrated to the residual proton and carbon resonance of the solvent: CDCl<sub>3</sub> ( $\delta_H$  7.25,  $\delta_C$  77.0 ppm), C<sub>6</sub>D<sub>6</sub> ( $\delta_H$  7.16,  $\delta_C$  128.0 ppm), CD<sub>3</sub>OD ( $\delta_H$  4.78, 3.21,  $\delta_C$  49.0 ppm), or [D<sub>6</sub>]acetone ( $\delta_H$  2.04,  $\delta_C$  29.8, 206.7 ppm). Conformational NMR analysis: Bruker Avance 600, recorded at 600.13 MHz proton resonance frequency at 300 K respectively 320 K (for 22) in  $[D_6]$ DMSO. Chemical shift calibration to  $[D_6]$ DMSO ( $\delta_H$  2.49 ppm,  $\delta_C$  39.5 ppm). Melting points: Büchi Melting Point B-540, uncorrected. IR: Jasco FT/IR-430. Optical rotation: Jasco polarimeter P-1020, reported in degree  $\lbrack a \rbrack_{D}$  (c  $\lbrack g$  per 100 mL], solvent). HRMS (FT-ICR): Bruker Daltonic APEX 2 with electron spray ionization (ESI). Flash chromatography: J. T. Baker silica gel 43-60 µm. Thin-layer chromatography Machery-Nagel Polygram Sil G/UV<sub>254</sub>. Analytical HPLC-MS: HP 1100 Series connected with an ESI MS detector Agilent G1946C, positive mode with fragmentor voltage of 40 eV, column: Nucleosil 100–5, C-18 HD, 5 µm, 70 × 3 mm Machery Nagel, eluent: NaCl solution (5 mm)/acetonitrile, gradient: 0/10/15/17/20 min with 20/80/80/99/ 99% acetonitrile, flow:  $0.6 \text{ mLmin}^{-1}$ . All solvents used in the reactions were purified before use. Dry diethyl ether, tetrahydrofuran, and toluene were distilled from sodium and benzophenone, whereas dry dichloromethane, dimethylformamide, pyridine, and triethylamine were distilled from CaH<sub>2</sub>. Petroleum ether with a boiling range of  $40-60^{\circ}$ C was used. The pH 7 buffer was prepared by dissolving  $KH_2PO_4$  (85.0 g, 0.625 mol) and NaOH (14.5 g, 0.3625 mol) in water (1 L). Reactions were generally run under an argon or nitrogen atmosphere. All commercially available compounds (Acros, Aldrich, Fluka, Merck) were used as received unless stated otherwise. L-Phenylalanine methyl ester and Boc-N-methyl-p-alanine  $(18)$  were prepared according to the literature reported.<sup>[16,17]</sup>

(2S)-3-(3-{(2S)-2-[(tert-Butoxycarbonyl)amino]propyl}phenyl)-2-methyl**propanoic acid (4):** NaOH (80 mg) in  $H_2O$  (5 mL) was added to a stirred solution of methyl ester 11 (0.55 g, 1.64 mmol) in THF (12 mL). The reaction mixture was stirred for 14 h at room temperature before being poured into water (25 mL) and extracted with diethyl ether  $(3 \times 10 \text{ mL})$ . The aqueous layer was acidified to pH 2–3 with 1n HCl and extracted with ethyl acetate  $(3 \times 15 \text{ mL})$ . The combined organic layers were dried  $(Na<sub>2</sub>SO<sub>4</sub>)$ , filtered, and concentrated. The crude product was purified by flash chromatography (ethyl acetate/petroleum ether 1:3) resulting in acid 4 as a white solid (0.44 g, 85%). M.p. 104-106 °C;  $R_f$  = 0.45 (ethyl acetate/petroleum ether 1:3);  $[\alpha]_D^{23} = +6.0$  (c=0.56, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta = 0.94$  (d,  $J = 6.1$  Hz, 3H; CH<sub>3</sub>NHR), 0.99 (d,  $J =$ 6.6 Hz, 3H; CH<sub>3</sub>CO<sub>2</sub>H), 1.28 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.45–2.66 (m, 4H; benzylic H, CH), 2.86 (dd,  $J=12.6, 6.1$  Hz, 1H; benzylic H), 3.19 (s, 1H; NH), 3.61–3.66 (m, 1H; CHNH), 6.91–6.92 (m, 3H; aryl H), 7.06 ppm (t, J= 7.5 Hz, 1 H; H<sub>m</sub>, aryl H); <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  = 17.2  $(CH_3CHNH)$ , 20.5 ( $CH_3CHCO_2H$ ), 28.8 ( $C(CH_3)_3$ ), 40.7 ( $CH_2CHCO_2H$ ), 42.7 (CHCO<sub>2</sub>H), 43.8 (CH<sub>2</sub>CHNH), 49.5 (CHNH), 79.8 (Boc C), 127.9, 128.3, 129.2, 131.2, 140.3, 140.7 (aryl), 157.7 (Boc C=O), 179.9 ppm  $(CO<sub>2</sub>H)$ ; IR (film):  $\tilde{\nu} = 3326$ , 2974, 2930, 1706, 1653, 1507, 1248, 1169 cm<sup>-1</sup>; HRMS (ESI):  $m/z$ : calcd for C<sub>18</sub>H<sub>27</sub>NO<sub>4</sub>Na: 344.18323; found: 344.18313 [M+Na]<sup>+</sup>.

(4R)-4-Benzyl-3-[(2S)-3-(3-{(2S)-3-[(4R)-4-benzyl-2-oxo-1,3-oxazolidin-3 yl]-2-methyl-3-oxopropyl}phenyl)-2-methylpropanoyl]-1,3-oxazolidin-2-

one (7): n-Butyllithium (22.5 mL, 56.2 mmol, 2.5m in hexane) was added at  $0^{\circ}$ C to a solution of diisopropylamine (8.0 mL, 56.2 mmol) in THF (190 mL). The reaction mixture was stirred at  $0^{\circ}$ C for 30 min, before it was cooled to  $-78$ °C. At this point, propionyl oxazolidinone 5 (12.0 g, 51.5 mmol), dissolved in THF (210 mL), was added. After being stirred for 1.5 h at  $-78^{\circ}$ C, the solid 1,3-bis(bromomethyl)-benzene (6) (6.17 g, 23.4 mmol) was added in one portion. Stirring was continued for 24 h with simultaneous warming of the reaction mixture to room temperature.

The reaction was quenched with  $NH<sub>4</sub>Cl$  (60 mL), and then most of the organic solvent was removed in vacuo. The residue was extracted with ethyl acetate  $(3 \times 50 \text{ mL})$  and the combined organic layers were washed with brine, dried with Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The residue was purified by flash chromatography (petroleum ether/EtOAc 7:3) to get 7 as a hygroscopic compound (7.72 g, 58%).  $R_f$  = 0.45 (petroleum ether/EtOAc 7:3);  $\left[\alpha\right]_D^{25} = -21.9$   $(c=1.503, \text{ CH}_2\text{Cl}_2)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.18 (d, J = 6.6 Hz, 6 H; CH<sub>3</sub>), 2.54–2.69 (m, 4 H; benzylic H), 3.11-3.18 (m, 4H; PhCH<sub>2</sub>), 4.10-4.20 (m, 6H; CHCH<sub>3</sub>, OCH2), 4.64–4.70 (m, 2H; NCH), 7.09–7.32 ppm (m, 14H; aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 16.2$  (CH<sub>3</sub>), 37.3 (PhCH<sub>2</sub>), 39.1 (CHCH<sub>3</sub>), 39.9 (benzylic), 54.7 (NCH), 65.5 (OCH<sub>2</sub>), 126.8, 127.9, 128.5, 129.0, 130.1, 134.9, 138.9 (aryl), 152.6 (NCO<sub>2</sub>), 176.1 ppm (CO); IR (film):  $\tilde{v} = 3028$ , 2976, 2932, 1770, 1694, 1604, 1588, 1487, 1455, 1393, 1288, 1210, 1103, 1053 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>34</sub>H<sub>36</sub>N<sub>2</sub>O<sub>6</sub>: 568.257798; found: 568.257289 [M]<sup>+</sup>.

(2S)-3-{3-[(2S)-2-Carboxypropyl]phenyl}-2-methylpropanoic acid (8):  $H<sub>2</sub>O<sub>2</sub>$  (11.7 mL of a 30 wt% solution, 102.7 mmol) was added at 0°C through a syringe to a solution of the bisalkylated compound 7 (7.50 g, 13.2 mmol) in THF (250 mL), followed by the addition of LiOH·H<sub>2</sub>O  $(2.20 \text{ g}, 51.4 \text{ mmol})$ , dissolved in water  $(120 \text{ mL})$ . The solution was stirred at 0°C for 5 h. Subsequently, saturated  $Na<sub>2</sub>SO<sub>3</sub>$  solution (100 mL) and saturated NaHCO<sub>3</sub> solution (100 mL) were added at  $0^{\circ}$ C. The whole mixture was partially concentrated in vacuo and diluted with water (100 mL). The aqueous layer was extracted with dichloromethane  $(3 \times$ 75 mL) to recover the auxiliary. The aqueous layer was then acidified at  $0^{\circ}$ C to pH 1.5 by using 6 M HCl and then extracted with ethyl acetate (4  $\times$  $100 \text{ mL}$ ). The combined organic layers were dried (MgSO<sub>4</sub>), filtered, and concentrated in vacuo yielding an oily residue (2.95 g, 90%).  $\lbrack a \rbrack_D^{25} =$  $+35.5$  (c=0.42, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$ =1.20 (d, J= 6.8 Hz, 6H; CH3), 2.62–2.72 (m, 2H; benzylic H), 2.75–2.83 (m, 2H; CH), 3.10 (dd,  $J=13.1$ , 6.1 Hz, 2H; benzylic H), 7.07 (d,  $J=7.1$  Hz, aryl H), 7.10 (s, 1H; H<sub>o</sub>, aryl H), 7.26 (t,  $J=7.6$  Hz, 1H; H<sub>m</sub>, aryl H), 11.79 ppm (brs, 2H; CO<sub>2</sub>H); <sup>13</sup>C NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 16.3$ (CH<sub>3</sub>), 39.1 (benzylic), 41.3 (CH), 127.1, 128.4, 129.7, 139.0 (aryl), 182.7 ppm (CO<sub>2</sub>H); IR (film):  $\tilde{v} = 3500-2500$  (broad), 1702, 1589, 1463, 1292, 1199, 1044 cm<sup>-1</sup>; HPLC-MS (ESI): m/z: 250.2, 204.2, 186.2, 177.2, 171.2, 131.2; HRMS (EI):  $m/z$ : calcd for C<sub>14</sub>H<sub>18</sub>O<sub>4</sub>: 250.12049; found:  $250.118291$   $[M]$ <sup>+</sup>.

Methyl (2S)-3-{3-[(2S)-3-methoxy-2-methyl-3-oxopropyl]phenyl}-2-meth**ylpropanoate** (9): A solution of DCC (8.12 g, 39.4 mmol) in  $CH_2Cl_2$ (26 mL) was added at  $0^{\circ}$ C to a solution of diacid 8 (3.20 g, 12.8 mmol), methanol (1.4 mL, 32.8 mmol) and DMAP (96 mg) in dry  $CH_2Cl_2$ (37 mL). The solution was stirred at  $0^{\circ}$ C for 30 min and then at room temperature for 6 h. The white precipitate was filtered off, the solvent evaporated, and the residue redissolved in diethyl ether. The organic solution was washed successively with cold  $1 \text{ N HCl}$ , NaHCO<sub>3</sub> solution, and brine. The dried  $(Na_2SO_4)$  organic layer was filtered, and concentrated. The crude product was purified by flash chromatography (ethyl acetate/ petroleum ether 1:9) to provide the diester 9 as an oily compound (2.52 g, 71%).  $R_f = 0.38$  (ethyl acetate/petroleum ether 1:9);  $[\alpha]_D^{25} = +40.1$  $(c=0.61, \text{ CH}_2\text{Cl}_2)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.15 (d, J = 6.5 Hz, 6H; CH3), 2.61–2.67 (m, 2H; benzylic H), 2.69–2.77 (m, 2H; CH), 3.00 (dd,  $J=13.1, 6.7 \text{ Hz}$ , 2H; benzylic H), 3.64 (s, 6H; OCH<sub>3</sub>), 6.97 (s, 1H;  $H_o$ , aryl H), 7.01 (d, J = 7.6 Hz, 2H; aryl H), 7.20 ppm (t, J = 7.6 Hz, 1H; H<sub>m</sub>, aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 16.6 (CH<sub>3</sub>), 39.6 (benzylic), 41.6 (CH), 51.5 (OCH3), 126.9, 128.3, 129.6, 139.3 (aryl), 176.4 ppm (CO<sub>2</sub>Me); IR (film):  $\tilde{v} = 2974$ , 2951, 1736, 1459, 1375, 1361, 1164 cm<sup>-1</sup>; HPLC-MS (ESI): m/z: 278.2, 218.2, 186.2, 171.2, 131.2, 105.2, 91.2; HRMS (EI):  $m/z$ : calcd for C<sub>16</sub>H<sub>22</sub>O<sub>4</sub>: 278.15179; found: 278.152648 [M]<sup>+</sup>.

(2S)-3-{3-[(2S)-3-Methoxy-2-methyl-3-oxopropyl]phenyl}-2-methylpropanoic acid (10): A solution of diester 9 (2.50 g, 9.0 mmol) in MeOH (5 mL) was emulsified under vigorous stirring in NaCl solution (0.1m, 646.75 mL) to which pH 7 phosphate buffer (3.25 mL) was added, making the solution 3 mm in phosphate. Then a suspension of PLE (25 mg, 1000 units, Sigma Aldrich, E-3019) in  $3.2 \text{M}$  (NH<sub>4</sub>)<sub>2</sub>SO<sub>4</sub> solution (1 mL) was added. During the hydrolysis the pH was kept between 7 and 7.5 by the controlled addition of NaOH solution (0.1n). After observing the formation of diacid in HPLC-MS (maximum 12 h), the reaction mixture was washed with CH<sub>2</sub>Cl<sub>2</sub> ( $2 \times 500$  mL). The aqueous phase was acidified to pH 2.5 with 25% hydrochloric acid and extracted with ethyl acetate  $(3 \times 500 \text{ mL})$ . The combined organic layers  $(CH_2Cl_2$  and EtOAc) were dried with Na<sub>2</sub>SO<sub>4</sub>, filtered, and concentrated in vacuo. The crude product was purified by flash chromatography (ethyl acetate/petroleum ether 1:4) to provide the monoacid monoester as an oily compound (1.52 g, 64%).  $R_f = 0.4$  (ethyl acetate/petroleum ether 1:4);  $[\alpha]_D^{25} = +46.15$  $(c=1.07, \text{ CH}_2\text{Cl}_2)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.05 (d, J = 6.82 Hz, 3H; CH<sub>3</sub>CHCO<sub>2</sub>CH<sub>3</sub>), 1.08 (d, J = 7.1 Hz, 3H; CH<sub>3</sub>CHCO<sub>2</sub>H), 2.53-2.58 (m, 2H; benzylic H), 2.62–2.68 (m, 2H; CH), 2.94 (ddd, J=19.4, 13.1, 6.4 Hz, 2H; benzylic H), 3.55 (s, 3H; OCH3), 6.90–6.96 (m, 3H; aryl H), 7.12 (t,  $J=7.5$  Hz, 1H; H<sub>m</sub>, aryl H), 10.49 ppm (broad, 1H; CO<sub>2</sub>H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 16.4 (CH<sub>3</sub>CO<sub>2</sub>CH<sub>3</sub>), 16.6 (CH<sub>3</sub>CO<sub>2</sub>H), 39.1 (CH<sub>2</sub>CHCO<sub>2</sub>CH<sub>3</sub>), 39.6 (CH<sub>2</sub>CHCO<sub>2</sub>H), 41.4 (CHCO<sub>2</sub>H), 51.5  $(OCH_3)$ , 126.9, 127.0, 128.4, 129.8, 139.0 (aryl), 176.4 (CO<sub>2</sub>Me), 182.3 ppm (CO<sub>2</sub>H); IR (film):  $\tilde{v} = 3025$ , 2975, 1736, 1707 cm<sup>-1</sup>; HPLC-MS: m/z: 264.2, 218.2, 186.2, 171.2, 158.2, 131.2, 91.2; HRMS (EI): m/z: calcd for  $C_{15}H_{20}O_4$ : 264.133877; found: 264.136141 [M]<sup>+</sup>.

Methyl (2S)-3-(3-{(2S)-2-[(tert-butoxycarbonyl)amino]propyl}phenyl)-2 methylpropanoate (11): A solution of the monoester 10 (0.80 g, 3.03 mmol) in toluene (23 mL) was treated with triethylamine (0.5 mL, 3.33 mmol) and DPPA (0.66 mL, 3.03 mmol). After stirring for 30 min, the mixture was heated under reflux for 3.5 h. The isocyanate formation was monitored by IR for the appearance of a strong signal in the 2300–  $2200 \text{ cm}^{-1}$  region and disappearance of the carboxylic acid carbonyl peak. The reaction mixture was cooled to 50 $\degree$ C, tert-butanol (3 mL, 10 equiv) was added via syringe and the solution heated to reflux for 20 h. The reaction was cooled to room temperature and quenched with saturated NaHCO<sub>2</sub> solution  $(25 \text{ mL})$ . The mixture was extracted with diethyl ether  $(3 \times 25 \text{ mL})$ . The combined organic extracts were dried  $(Na_3SO_4)$ , filtered, and concentrated in vacuo. The crude product was purified by flash chromatogrophy (ethyl acetate/petroleum ether 1:5), to provide compound 11 as an oil (0.73 g, 72%).  $R_f$  = 0.55 (ethyl acetate/petroleum ether 1:5);  $\left[\alpha\right]_D^{23} = +12.3$  (c=0.21, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 0.99$  (d,  $J = 6.6$  Hz, 3H; CH<sub>3</sub>CHCO<sub>2</sub>Me), 1.06 (d,  $J = 6.6$  Hz, 3H; CH<sub>3</sub>CHNR), 1.36 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.51–2.58 (m, 2H; benzylic H), 2.63-2.68 (m, 1H; CHCO<sub>2</sub>Me), 2.75 (dd, J=13.1, 5.3 Hz, 1H; benzylic H), 2.93 (dd,  $J=13.1$ , 6.4 Hz, 1H; benzylic H), 3.57 (s, 3H; OCH<sub>3</sub>), 3.81 (broad, 1H; CHNHR), 4.31 (broad, 1H; NH), 6.90–7.32 ppm (m, 4H; aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 16.3$  (CH<sub>3</sub>CHCO<sub>2</sub>Me), 19.7  $(CH_3CHNHR)$ , 28.0  $(C(CH_3)_3)$ , 39.2  $(CH_2CHCO_2Me)$ , 41.0  $(CHCO<sub>2</sub>Me<sub>3</sub>), 42.5 (CH<sub>2</sub>CHNHR), 47.1 (CHNHR), 51.2 (OCH<sub>3</sub>), 78.8)$ (Boc quaternary C), 126.6, 127.2, 127.9, 129.9, 137.9, 139.0 (aryl), 154.8 (Boc CO), 176.2 ppm (CO<sub>2</sub>Me); IR (film):  $\tilde{v} = 3361, 2973, 2930, 2359,$ 1735, 1710, 15516, 1364, 1166 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for  $C_{19}H_{29}NO_4$ : 335.209636; found: 335.207186  $[M]$ <sup>+</sup>.

Methyl N-(tert-butoxycarbonyl)-D-alanyl-L-phenylalaninate (14): A solution of DCC (1.5 g, 7.25 mmol) in THF (11 mL) at  $0^{\circ}$ C was added to a stirred solution of Boc-D-Ala-OH (13) (1.10 g, 5.6 mmol), H-L-Phe-OMe (12)  $(1.00 \text{ g}, 5.6 \text{ mmol})$ , hydroxybenzotriazole  $(0.75 \text{ g}, 5.6 \text{ mmol})$  in dry THF (45 mL). Stirring was continued for 12 h at room temperature. The dicyclohexylurea was filtered off, washed with cold diethyl ether, and the filtrate was concentrated. Purification was done by flash chromatography (ethyl acetate/petroleum ether 1:3) yielding a colorless solid (1.55 g, 80%). M.p. 98–99 °C;  $R_f = 0.35$  (ethyl acetate/petroleum ether 1:3);  $[\alpha]_{\text{D}}^{23}$  = +59.8 (c = 0.40, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.21 (d,  $J=7.1$  Hz, 3H; CH<sub>3</sub>), 1.36 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 3.00 (dd,  $J=13.8$ , 6.2 Hz, 1H; benzylic H), 3.05–3.10 (m, 1H; benzylic H), 4.04–4.18 (m, 1H; CHNH), 4.77-4.81 (m, 1H; CHCO<sub>2</sub>Me), 4.95 (brs, 1H; NHBoc), 6.60 (brs, 1H; NHCHCO<sub>2</sub>Me), 7.03-7.22 ppm (m, 5H; aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 18.4$  (CH<sub>3</sub>), 28.2 (C(CH<sub>3</sub>)<sub>3</sub>), 37.8 (benzylic), 49.9 (CHCO2Me), 52.3 (OCH3), 53.0 (CHNHBoc), 80.0 (Boc C), 127.1, 128.5, 129.2, 135.7 (aryl), 155.3 (Boc C=O), 171.7 (CO<sub>2</sub>Me), 172.2 ppm (CONH); IR (KBr):  $\tilde{v} = 3304$ , 2987, 2930, 1732, 1664, 1517, 1317 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>14</sub>H<sub>18</sub>N<sub>2</sub>O<sub>5</sub>: 294.119319; found: 294.121541  $[M-C(CH<sub>3</sub>)<sub>3</sub>$ <sup>+</sup>.

Methyl N-(tert-Butoxycarbonyl)-L-alanyl-p-alanyl-L-phenylalaninate (15): A solution of peptide 14 (1.0 g, 2.8 mmol) in  $CH_2Cl_2$  (22 mL) was treated with  $CF<sub>3</sub>CO<sub>2</sub>H$  (2.2 mL) and the mixture stirred at room temperature for 1 h. The solvent was removed in vacuo, and the residue dried by azeotropic removal of  $H_2O$  with toluene. The crude material was subjected to the next reaction without further purification. To a cooled  $(0^{\circ}C)$  solution of the crude amine salt (240 mg, 0.96 mmol) and Boc-l-Ala-OH (181 mg, 0.96 mmol) in THF (11 mL) and CH<sub>2</sub>Cl<sub>2</sub> (2.5 mL) were added 1-hydroxybenzotriazole (130 mg, 0.96 mmol), Et<sub>3</sub>N (0.32 mL, 2.3 mmol), and 1-(3-<br>dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDCI) dimethylaminopropyl)-3-ethylcarbodiimide hydrochloride (EDCI) (239 mg, 1.25 mmol). The mixture was then stirred at room temperature for 16 h. The solvent was removed in vacuo, and the residue purified by flash chromatography (40% ethyl acetate in petroleum ether) to provide tripeptide 15 as a colorless solid (0.30 g, 75%). M.p. 149–151 °C;  $R_f$  = 0.33 (40% ethyl acetate in petroleum ether);  $[\alpha]_D^{28} = +19.1$  ( $c = 0.48$ , CH<sub>2</sub>Cl<sub>2</sub>);<br><sup>1</sup>H NMP (400 MHz CDCL);  $\lambda = 1.19$  (d  $I = 7.1$  Hz 3H; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 1.19$  (d,  $J = 7.1$  Hz, 3H; CH<sub>3</sub>CHNHBoc), 1.24 (d,  $J=7.0$  Hz, 3H; CH<sub>3</sub>CONH), 1.36 (s, 9H; C- $(CH_3)$ , 2.96 (dd, J=13.9, 7.1 Hz, 1H; benzylic H), 3.07–3.12 (m, 1H; benzylic H), 3.62 (s, 3H; OCH3), 4.02–4.17 (m, 1H; CHNHBoc), 4.40– 4.47 (m, 1H; CHNHCO), 4.73-4.78 (m, 1H; CHCO<sub>2</sub>Me), 5.11 (d,  $J=$ 7.3 Hz, 1H; NHBoc), 6.84 (d,  $J=7.3$  Hz, 1H; NHCO<sub>2</sub>Me), 6.95 (d,  $J=$ 6.6 Hz, 1H; NHCO), 7.05–7.22 ppm (m, 5H; aryl H); 13C NMR (100 MHz, CDCl<sub>3</sub>): 18.1 (CH<sub>3</sub>CHNHBoc), 18.4 (CH<sub>3</sub>CHCONH), 28.3 C- $(CH<sub>3</sub>)$ , 37.8 (benzylic), 48.5 (CHCONH), 50.2 (CHNHBoc), 53.2 (CHCO2Me), 80.1 (Boc C), 127.0, 128.5, 129.2, 135.8 (aryl), 155.3 (Boc C=O), 171.7 (CO<sub>2</sub>Me), 171.8 (NHCO), 172.6 ppm (CH<sub>2</sub>NHCO); IR (KBr):  $\tilde{v} = 3277$ , 2979, 2748, 1753, 1707, 1645, 1519, 1456, 1370, 1166 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>21</sub>H<sub>31</sub>N<sub>3</sub>O<sub>6</sub>: 421.221247; found: 421.225247 [M] +.

Methyl N-[(2S)-3-(3-{(2S)-2-[(tert-butoxycarbonyl)amino]propyl}phenyl)- 2-methylpropanoyl]-L-alanyl-D-alanyl-L-phenylalaninate (16): A solution of peptide 15 (100 mg, 0.24 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (2 mL) was treated with  $CF<sub>3</sub>CO<sub>2</sub>H$  (0.18 mL), and the mixture stirred at room temperature for 1 h. The solvent was removed in vacuo, and the residue was dried by azeotropic removal of H<sub>2</sub>O with toluene. The crude material was subjected to the next reaction without further purification. To a solution of crude amine salt and Boc-protected w-amino acid 4 (77 mg, 0.24 mmol) in DMF (2 mL) were added TBTU (77 mg, 0.24 mmol) HOBt (34 mg, 0.24 mmol), DIEA (0.1 mL, 0.576 mmol) and the mixture was stirred for 3 h at room temperature. The reaction mixture was diluted with water (3 mL) and extracted with ethyl acetate  $(3 \times 4$  mL). The combined organic layers were washed with water resulting in almost pure tetrapeptide 16 (133 mg, 90%) as judged by HPLC-MS. This material was used for the macrolactam formation without any further purification. N-Boc-Nmethyl-n-alanine (18) was prepared according to the literature.<sup>[23,24]</sup>

#### (3S,6S,9R,12S,15S)-6-Benzyl-3,9,12,15-tetramethyl-4,7,10,13-

tetraazabicyclo[15.3.1]henicosa-1(21),17,19-triene-5,8,11,14-tetrone (17) (deprotection of the carboxylic and amino groups of compound 16 and macrocyclization): NaOH  $(7.7 \text{ mg})$ , dissolved in H<sub>2</sub>O  $(0.5 \text{ mL})$ , was added to a stirred solution of tetrapeptide 16 (100 mg, 0.160 mmol) in THF (3 mL). The reaction mixture was stirred for 1 h at room temperature before being poured into water (5 mL) and extracted with diethyl ether ( $3 \times 5$  mL). The aqueous layer was acidified to pH 2–3 with 1n HCl and extracted with ethyl acetate  $(3 \times 5 \text{ mL})$ . The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and evaporated providing the carboxylic acid in almost quantitative yield. This compound was used directly in the next step. A solution of above Boc-protected tetrapeptide acid (90 mg, 0.145 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (1.2 mL) was treated with CF<sub>3</sub>CO<sub>2</sub>H (0.11 mL, 1.45 mmol), and the mixture stirred at room temperature for 1 h. The solvent was removed in vacuo, and the residue dried by azeotropic removal of H2O with toluene. The crude material was subjected to the macrolactamization without any further purification. Thus, the residue was dissolved in DMF (140 mL) and the stirred solution treated successively with TBTU (140 mg, 0.435 mmol), HOBt (59 mg, 0.435 mmol), and  $iPr<sub>2</sub>EtN$  (0.08 mL, 0.44 mmol) at room temperature. The resulting solution was stirred for 14 h at room temperature and then partitioned between ethyl acetate and water. After separation of the layers, the aqueous layer was extracted with ethyl acetate  $(2 \times 75 \text{ mL})$ , and the combined organic layers were washed successively with water, 5% aqueous

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 $KHSO<sub>4</sub>$ , water, half-saturated NaHCO<sub>3</sub>, and brine. After being dried (MgSO4), filtered, and concentrated in vacuo, the residue was purified by flash chromatography (ethyl acetate) to give the macrocycle 17 as a colorless solid (40 mg, 50%, three steps). M.p. 258-260°C;  $R_f$ =0.5 (ethyl acetate);  $[\alpha]_D^{28} = -14.8$   $(c=0.19, \text{ CH}_2\text{Cl}_2)$ ; <sup>1</sup>H NMR (600 MHz, [ $D_6$ ]DMSO):  $\delta$  = 1.01 (d, J = 7.3 Hz, 5-CH<sub>3</sub>), 1.06 (d, J = 6.6 Hz, 3H; 11-CH<sub>3</sub>), 1.11 (d,  $J=7.3$  Hz, 3H; 5-CH<sub>3</sub>), 1.12 (d,  $J=6.6$  Hz, 3H; 14-CH<sub>3</sub>), 2.50 (m, 2H; 6-H, 11-H), 2.59–2.67 (m, 2H; 10-H), 2.86–2.95 (m, 2H; 1- H, 6-H), 3.33 (dd, J=13.9, 3.7 Hz, 1H; 1-H), 3.74–3.79 (m, 1H; 17-H), 3.94–4.03 (m, 3H; 2-H, 14-H, 5-H), 6.75–6.79 (m, 2H; 4''-H, 6''-H), 6.83 (d,  $J=7.3$  Hz, 2H; 4-NH, 13-NH), 6.94 (s, 1H; 2"-H), 6.98 (dd,  $J=7.3$ , 7.3 Hz, 1H; 5''-H), 7.15–7.19 (m, 3H; 4'-H, aryl H), 7.23–7.27 (m, 2H; aryl H), 8.17 (d, J=8.1 Hz, 1H; 19-NH), 8.34 ppm (brs, 1H; 16-NH); <sup>13</sup>C NMR (150 MHz,  $[D_6]$ DMSO):  $\delta$  = 15.8 (CH<sub>3</sub>-17), 17.9 (CH<sub>3</sub>-11), 18.3, 19.1 (CH<sub>3</sub>-5, CH<sub>3</sub>-14), 34.6 (C-1), 39.1 (C-6), 39.5 (C-10), 42.0 (C-11), 45.0 (C-5), 47.6 (C-14), 49.6 (C-17), 54.4 ppm (C-2); IR (film):  $\tilde{v} = 3297$ , 2931, 1888, 1640,1526, 1446 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for  $C_{28}H_{36}N_4O_4$ : 492.273621; found: 492.271404  $[M]^+$ .

Methyl N-(tert-butoxycarbonyl)-N-methyl-D-alanyl-L-phenylalaninate (19): A solution of DCC (2.3 g, 11.1 mmol), dissolved in THF (11 mL) was added at  $0^{\circ}$ C to a stirred solution of Boc-N-Me-D-Ala-OH (18) (1.5 g, 7.4 mmol), H-l-Phe-OMe (12) (1.3 g, 7.4 mmol), hydroxybenzotriazole (0.99 g, 7.4 mmol) in dry THF (60 mL). Stirring was continued for 7 h at room temperature. The dicyclohexylurea was filtered off, washed with cold diethyl ether, and the filtrate concentrated. Purification of the residue by flash chromatography (ethyl acetate/petroleum ether 1:5) gave a gel-like compound (2.10 g, 75%).  $R_f$ =0.33 (ethyl acetate/petroleum ether 1:5);  $\left[\alpha\right]_D^{25} = +62.8$  (c=0.89, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta =$ 1.23 (d,  $J=6.8$  Hz,  $3H$ ; CH<sub>3</sub>), 1.37 (s,  $9H$ ; C(CH<sub>3</sub>)<sub>3</sub>), 2.64 (s,  $3H$ ; NCH<sub>3</sub>), 2.99–3.07 (m, 2H; benzylic H), 3.63 (s, 3H; OCH3), 4.71–4.77 (m, 2H; CHCO<sub>2</sub>Me, CHMe), 6.50 (brs, 1H; NHCHCO<sub>2</sub>Me), 7.03 (d,  $J=7.3$  Hz, 2H; H<sub>o</sub>, aryl H), 7.15–7.23 ppm (m, 3H; aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 13.2$  (CH<sub>3</sub>), 28.3 (C(CH<sub>3</sub>)<sub>3</sub>), 29.7 (NCH<sub>3</sub>), 37.8 (benzylic), 52.2 (OCH3, CHCO2Me), 52.9 (CHNHBoc), 80.5 (Boc C), 127.1, 128.6, 129.1, 135.7, 171.3 (CO<sub>2</sub>Me), 171.7 ppm (CONH); IR (film):  $\tilde{v} = 3327$ , 2977, 2934, 2118, 1746, 1688, 1515, 1455, 1390, 1154 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>19</sub>H<sub>28</sub>N<sub>2</sub>O<sub>5</sub>: 364.199791, found 364.198514 [M]<sup>+</sup>.

Methyl N-(tert-Butoxycarbonyl)-L-alanyl-N-methyl-D-alanyl-L-phenylalaninate (20): A solution of peptide 19 (1.0 g, 2.8 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (22 mL) was treated with  $CF_3CO_2H$  (2.2 mL), and the mixture was stirred at room temperature for 1 h. The solvent was removed in vacuo, and the residue dried by azeotropic removal of  $H_2O$  with toluene. The crude material was subjected to the next reaction without further purification.

To a stirred solution of the crude amine salt, Boc-L-Ala-OH (0.53 g, 2.8 mmol), PyBroP  $(1.3 g, 2.8 mmol)$  in CH<sub>2</sub>Cl<sub>2</sub>  $(3 mL)$  was added DIPEA (1.4 mL, 8.4 mmol) at  $0^{\circ}$ C and the mixture stirred at room temperature for 3 h. The solvent was removed in vacuo and the residue purified by flash chromatography (ethyl acetate/petroleum ether 1:1) to provide a gel-like compound (0.69 g, 58%).  $R_f$  = 0.45 (ethyl acetate/petroleum ether 1:1);  $[\alpha]_D^{25} = +62.4$  (c=1.79, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 1.21$  (d,  $J = 6.8$  Hz, 3H; CH<sub>3</sub>CHNCH<sub>3</sub>), 1.23 (d,  $J = 7.8$  Hz, 3H; CH<sub>3</sub>CHNHBoc), 1.37 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.85 (s, 3H; NCH<sub>3</sub>), 2.97 (dd, J=13.9, 7.1 Hz, 1H; benzylic H), 3.04–3.09 (m, 1H; benzylic H), 3.60 (s, 3H; OCH3), 4.44–4.50 (m, 1H; CHNHBoc), 4.65–4.70 (m, 1H; CHCO<sub>2</sub>Me), 5.11 (q,  $J=6.3$  Hz, 1H; CHNCH<sub>3</sub>), 5.31 (d,  $J=6.8$ , 1H; NHBoc), 6.70 (d,  $J=7.8$  Hz, 1H; NHCHCO<sub>2</sub>Me), 7.05 (d,  $J=7.3$ , 2H; H<sub>o</sub>, aryl H), 7.14–7.24 ppm (m, 3H; aryl H); <sup>13</sup>C NMR (100 MHz,  $CDCl<sub>3</sub>$ ): 13.5 (CH<sub>3</sub>CHCO), 17.9 (CH<sub>3</sub>CHNHBoc), 28.2 (C(CH<sub>3</sub>)<sub>3</sub>), 30.3 (NCH<sub>3</sub>), 37.4 (benzylic), 46.6 (CHNHBoc), 52.1 (OCH<sub>3</sub>, CHNMe), 53.0 (CHCO2Me), 79.6 (Boc C), 127.0, 128.4, 129.0, 135.9 (aryl), 155.3 (Boc C=O), 170.4 (CO2Me), 171.7 (CONMe), 173.7 ppm (CONH); IR (film):  $\tilde{v} = 3327, 2979, 1742, 1682, 1642, 1520, 1455, 1249, 1169 \text{ cm}^{-1}; \text{ HRMS}$ (EI):  $m/z$ : calcd for C<sub>22</sub>H<sub>33</sub>N<sub>3</sub>O<sub>6</sub>: 435.236897; found: 435.240391 [M]<sup>+</sup>.

Methyl N-[(2S)-3-(3-{(2S)-2-[(tert-butoxycarbonyl)amino]propyl}phenyl)- 2-methylpropanoyl]-L-alanyl N-methyl-D-alanyl-L-phenylalaninate (21): A solution of tripeptide  $20$  (180 mg, 0.413 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (3.5 mL) was treated with  $CF_3CO_2H$  (0.32 mL), and the mixture stirred at room temperature for 1 h. The solvent was removed in vacuo, and the residue

dried by azeotropic removal of H<sub>2</sub>O with toluene. The crude material was subjected to the next reaction without further purification.

To a cooled  $(0^{\circ}C)$  solution of the crude amine salt and amino acid 4 (133 mg, 0.413 mmol) in THF (7 mL) and  $CH_2Cl_2$  (1.5 mL) were added 1-hydroxybenzotriazole (56.2 mg, 0.413 mmol),  $Et_3N$  (0.15 mL, 1.03 mmol), and EDCI (103 mg, 0.54 mmol), followed by stirring of the mixture at room temperature for 16 h. The solvent was removed in vacuo, and the residue purified by flash chromatography (ethyl acetate/ petroleum ether 1:1) to provide a gel-like compound (0.14 g, 55%).  $R_f$ = 0.33 (ethyl acetate/petroleum ether 1:1);  $\left[\alpha\right]_D^{25} = +61.5$  (c=0.52, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CD<sub>3</sub>OD):  $\delta$  = 1.04 (d, J = 6.6, 6H; CH<sub>3</sub>CHNH, CH<sub>2</sub>CH(CH<sub>3</sub>)), 1.23 (d, J=7.3 Hz, 3H; CH<sub>3</sub>CHN(CH<sub>3</sub>)), 1.30 (d, J= 7.1 Hz, 3H; CH<sub>3</sub>CHNHBoc), 1.36 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.40–2.72 (m, 4H; benzylic H), 2.86 (s, 3H; NCH<sub>3</sub>), 3.09–3.16 (m, 2H; PhCH<sub>2</sub>), 3.58 (s, 3H; OCH<sub>3</sub>), 3.68 (m, 1H; NHBoc), 4.50-4.60 (m, 2H; CH(CH<sub>3</sub>)NH, CH- $(CH_3)NHBoc$ ), 5.03 (q, J = 7.1 Hz, 1H; CH(CH<sub>3</sub>)NCH<sub>3</sub>), 6.89–7.20 (m, 11H; aryl H,  $PhCH_2CHNH$ ), 7.90 ppm (broad, 1H; COCHNH); <sup>13</sup>C NMR (100 MHz, CD<sub>3</sub>OD):  $\delta$  = 14.0 (CH<sub>3</sub>CHN(CH<sub>3</sub>)), 16.7 (CH<sub>2</sub>CH- $(CH_3)$ ), 17.4 (CH<sub>3</sub>CHNH), 20.6 (CH<sub>3</sub>CHNHBoc), 28.8 (C(CH<sub>3</sub>)<sub>3</sub>), 31.4  $(NCH<sub>3</sub>), 38.1$  (PhCH<sub>2</sub>), 38.9 (CH(CH<sub>3</sub>)CO), 40.7 (CH<sub>2</sub>CH(CH<sub>3</sub>)CO), 43.0  $(CH_2CH(CH_3)NH)$ , 47.3 ( $CH(CH_3)NH)$ , 52.7 ( $CH(CH_3)NCH_3$ ), 53.8 (OCH<sub>3</sub>), 55.51 (CHCO<sub>2</sub>Me), 79.8 (Boc C), 127.8, 128.0, 128.3, 129.2, 129.4, 130.2, 138.4, 140.4, 140.8 (aryl), 157.7 (Boc C=O), 173.1 (N-  $(CH_3)CO$ ), 173.4  $(CO_2Me)$ , 175.2  $(COCHN(CH_3))$ , 178.7 ppm (NHCOCH); IR (film):  $\tilde{v} = 3315$ , 2975, 2932, 1742, 1644, 1526, 1455, 1391, 1247, 1172 cm<sup>-1</sup>; HRMS (ESI):  $m/z$ : calcd for C<sub>35</sub>H<sub>50</sub>N<sub>4</sub>O<sub>7</sub>Na: 661.35717; found: 661.34712 [M+Na]<sup>+</sup>.

(3S,6S,9R,12S,15S)-6-Benzyl-3,9,10,12,1-pentamethyl-4,7,10,13-tetraazabicyclo[15.3.1]henicosa-1,17,19-trien-5,8,11,14-tetrone (22): NaOH (7.5 mg) in  $H<sub>2</sub>O$  (0.5 mL) was added to a stirred solution of tetrapeptide 21 (100 mg, 0.156 mmol) in THF (3 mL). The reaction mixture was stirred for 1 h at room temperature before being poured into saturated  $NAHCO<sub>3</sub>$ solution (5 mL) and extracted with diethyl ether  $(3 \times 5$  mL). The aqueous layer was acidified to pH 2–3 with 1n HCl and extracted with ethyl acetate ( $3 \times 5$  mL). The combined organic layers were dried (Na<sub>2</sub>SO<sub>4</sub>), filtered, and concentrated in vacuo to give the free acid in almost quantitative yield. This acid was used for the next step without further purification. To a solution of above N-Boc protected tetrapeptide acid (90 mg, 0.144 mmol) in  $CH_2Cl_2$  (1.2 mL) was added  $CF_3CO_2H$  (0.11 mL, 1.45 mmol), and the mixture was stirred at room temperature for 1 h. The solvent was removed in vacuo, and the residue dried by azeotropic removal of H2O with toluene. The crude material was subjected to the macrolactamization without any further purification.

The residue was dissolved in DMF (140 mL) and the stirred solution was treated successively with TBTU (140 mg, 0.435 mmol), HOBt (59 mg, 0.435 mmol), and  $iPr_2NEt(0.08 \text{ mL}, 0.435 \text{ mmol})$  at room temperature. The solution was stirred at room temperature for 14 h and then partitioned between ethyl acetate and water. After separation of the layers, the aqueous layer was extracted with ethyl acetate  $(2 \times 75 \text{ mL})$ . The combined organic layers were washed successively with water, 5% aqueous  $KHSO<sub>4</sub>$ , water, half-saturated NaHCO<sub>3</sub> and brine, dried (MgSO<sub>4</sub>), filtered, and concentrated in vacuo. The residue was purified by flash chromatography (ethyl acetate) to provide a colorless sticky solid (50 mg, 62%, three steps).  $R_f = 0.55$  (ethyl acetate);  $[\alpha]_D^{28} = +18.0$  (c=0.30, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (600 MHz, [D<sub>6</sub>]DMSO):  $\delta = 0.97$  (d, J = 7.3 Hz, 17-CH<sub>3</sub>), 1.07 (d,  $J=7.3$  Hz, 3H; CH<sub>3</sub>), 1.09 (d,  $J=5.9$  Hz, 3H; 11-CH<sub>3</sub>), 1.10 (d,  $J=5.9$  Hz,  $3H$ ; CH<sub>3</sub>), 2.36–2.43 (m, 1H; 11-H), 2.48 (m, 1H; 10-H), 2.49 (m, 1H; 6-H), 2.60–2.70 (m, 2H; 10-H, 1-H), 2.84 (s, 3H; NCH<sub>3</sub>), 2.95 (dd,  $J=13.2$ , 3.7 Hz, 1H; 6-H), 3.35 (dd,  $J=13.6$ , 3.3 Hz, 1H; 1-H), 4.04–4.08 (m, 1H; 5-H), 4.22–4.29 (m, 2H; 17-H, 2-H), 4.39– 4.44 (m, 1H; 14-H), 6.46 (br s, 3H; 13-NH), 6.50 (d, J=8.1 Hz, 4-NH), 6.65, 6.70 (2 d,  $J=7.3$  Hz, 2 H; 4"-H, 6"-H), 6.90 (dd,  $J=7.3$ , 7.3 Hz, 1 H; 5''-H), 6.93 (s, 1H; 2''-H), 7.15–7.19 (m, 1H; 4'-H), 7.21–7.27 (m, 4H; 2'- H, 4'-H), 8.31 ppm (d,  $J=8.8$  Hz, 1H; 19-NH); <sup>13</sup>C NMR (150 MHz,  $[D_6]$ DMSO):  $\delta$  = 14.3 (CH<sub>3</sub>-17), 17.5 (CH<sub>3</sub>-11), 18.4 (CH<sub>3</sub>-5, CH<sub>3</sub>-14), 30.5 (NCH<sub>3</sub>), 35.4 (C-1), 38.6 (C-6), 40.5 (C-10), 43.6 (C-11), 44.3 (C-5), 44.7 (C-14), 53.3 (C-17), 53.4 ppm (C-2); IR (film):  $\tilde{v} = 3302$ , 2933, 1633, 1520,

1455 cm<sup>-1</sup>; HRMS (ESI):  $m/z$ : calcd for C<sub>29</sub>H<sub>38</sub>N<sub>4</sub>O<sub>4</sub>Na: 529.27853; found: 529.27827 [M+Na]<sup>+</sup>.

tert-Butyl (3R)-4-[(4R)-4-Benzyl-2-oxo-1,3-oxazolidin-3-yl]-3-(4-methoxy**phenyl)-4-oxobutanoate** (24): A solution of compound 23 (5.20 g, 0.016 mol) in anhydrous THF (32 mL) was treated at  $-78^{\circ}$ C with NaHMDS in THF (2M, 8.7 mL, 0.017 mol). After stirring for 2.5 h at the same temperature, tert-butyl bromoacetate (6.5 mL, 0.048 mol) was added and the mixture and stirred for additional 3 h at this temperature. The mixture was allowed to warm to  $0^{\circ}C$ , before saturated NH<sub>4</sub>Cl solution (100 mL) was added. Most of the THF was then removed on the rotary evaporator and the resulting slurry extracted with ethyl acetate  $(3 \times 90 \text{ mL})$ . The combined organic layers were dried with MgSO<sub>4</sub>, filtered, and concentrated to give the crude product as a yellow solid. Purification by flash chromatography (petroleum ether/ethyl acetate 4:1) and recrystallization from a mixture of hexane/ether gave the pure alkylation product as white needles (5.1 g, 71%). M.p. 139.5–140.5 °C;  $R_f$  = 0.23 (petroleum ether/ethyl acetate 4:1);  $[a]_{D}^{25} = +149.0$  (c=1.02, CH<sub>2</sub>Cl<sub>2</sub>);<br><sup>1</sup>H NMP (400 MHz CDCL);  $\delta = 1.36$  (c 0 H; C(CH)) 2.51 (dd L=17.1 <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.36 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.51 (dd, J = 17.1, 4.5 Hz, 1H; CH<sub>2</sub>CO), 2.72 (dd,  $J=13.3$ , 9.8 Hz, 1H; PhCH<sub>2</sub>), 3.19 (dd,  $J=16.9, 11.3$  Hz, 1H; CH<sub>2</sub>CO), 3.28 (dd,  $J=13.3, 2.7$  Hz, 1H; PhCH<sub>2</sub>), 3.69 (s, 3H; OCH<sub>3</sub>), 3.91 (t,  $J=8.0$  Hz, 1H; CH<sub>2</sub>O), 3.99–4.02 (m, 1H; CH2O), 4.47–4.52 (m, 1H; NCH), 5.36 (dd, J=11.3, 4.5 Hz, 1H; CHCO), 6.75 (d,  $J=8.5$  Hz, 2H; aryl H), 7.17–7.28 ppm (m, 7H; aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 27.9$  (C(CH<sub>3</sub>)<sub>3</sub>), 37.4 (PhCH<sub>2</sub>), 40.4 (CH<sub>2</sub>CO), 43.8 (CHCO), 55.1 (OCH<sub>3</sub>), 55.6 (NCH), 65.5 (CH<sub>2</sub>O), 80.7  $(C(CH_3)$ , 114.0, 127.1, 128.8, 128.9, 129.4, 129.5, 135.5, 152.6 (NCO<sub>2</sub>), 158.9 (aryl), 170.9 (CHCO), 173.4 ppm (CH<sub>2</sub>CO<sub>2</sub>); IR (film):  $\tilde{v} = 2356$ , 1697, 1646, 1519, 1045 cm<sup>-1</sup>; MS (EI):  $m/z$  (%): 383.1 (6), 206.0 (15), 178.1 (100), 83.9 (20); HRMS (EI):  $m/z$ : calcd for  $C_{21}H_{21}NO_6$ : 383.140500; found: 383.136857  $[M-C(CH_3)_3]^+$ .

(2R)-4-tert-Butoxy-2-(4-methoxyphenyl)-4-oxobutanoic acid (25): A solution of H<sub>2</sub>O<sub>2</sub> (30% in H<sub>2</sub>O<sub>2</sub> 8.3 mL, 0.069 mol) at 0<sup>o</sup>C was added dropwise to a solution of compound 24 (5.10 g, 0.0133 mol) in THF (196 mL) followed by a solution of LiOH in  $H_2O$  (0.3m in  $H_2O$ , 66 mL, 0.023 mol). The reaction mixture was stirred at the same temperature for 5 h, at which point TLC indicated the complete consumption of the starting material. Then saturated aqueous solutions of  $Na<sub>2</sub>SO<sub>3</sub>$  and  $Na<sub>2</sub>CO<sub>3</sub>$  (40 mL) of each) were added. The mixture was partly concentrated on the rotary evaporator to remove THF, then diluted with H<sub>2</sub>O (85 mL), and extracted with  $CH_2Cl_2$  (3×100 mL). The combined organic layers were dried with MgSO4, filtered, and concentrated in vacuo to afford recovered chiral auxiliary (1.5 g, 78% yield). The aqueous layer was acidified to pH 1.5–2.5 with 6m HCl at  $0^{\circ}$ C and extracted with ethyl acetate ( $3 \times 85$  mL). The combined organic layers were dried over anhydrous MgSO<sub>4</sub>, filtered, and concentrated in vacuo to provide the carboxylic acid 25 (2.5 g, 78%). This acid was used in the next step without further purification. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.38 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.56 (dd, J = 16.6, 5.5 Hz, 1H; CH<sub>2</sub>), 3.03 (dd,  $J=16.4$ , 9.8 Hz, 1H; CH<sub>2</sub>), 3.77 (s, 3H; OCH<sub>3</sub>), 3.99 (dd,  $J=10.1$ , 5.8 Hz, 1H; CH), 6.83 (d,  $J=8.8$  Hz, 2H; aryl H), 7.19 (d,  $J=8.5$  Hz, 2H; aryl H), 10.53 ppm (s, 1H; CO<sub>2</sub>H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 27.8$  (C(CH<sub>3</sub>)<sub>3</sub>), 38.6 (CH<sub>2</sub>), 46.4 (CH), 55.2 (OCH<sub>3</sub>), 81.1 (C- $(CH<sub>3</sub>)<sub>3</sub>$ ), 114.0, 128.9, 129.1, 159.0, 170.5, 179.2 ppm (CHCO<sub>2</sub>); IR (film):  $\tilde{v} = 2977$ , 1720, 1253, 1160 cm<sup>-1</sup>; HRMS (ESI):  $m/z$ : calcd for  $C_{15}H_{20}O_5Na$ : 303.12029; found: 303.11986  $[M+Na]^+$ .

tert-Butyl (3R)-3-{[(9H-fluoren-9-ylmethoxy)carbonyl]amino}-3-(4-methoxyphenyl)propanoate  $(26)$ : Et<sub>3</sub>N  $(0.5 \text{ mL}, 3.6 \text{ mmol})$  and DPPA (0.71 mL, 3.2 mmol) were added to a solution of carboxylic acid 25 (0.92 g, 3.2 mmol) in toluene (18 mL). The reaction mixture was stirred at room temperature for 30 min and then heated to reflux. Evolution of  $N_2$  was observed between 70–80 °C. The reaction progress was monitored by HPLC-MS. After complete conversion of the starting material to the isocyanate (3.5–4 h), the reaction mixture was cooled to  $50^{\circ}$ C and then treated carefully with FmocOH (1.93 g, 0.009 mol). The mixture was refluxed for 3 h. After the complete conversion of the isocyanate to the carbamate 26, the mixture was cooled to room temperature and diluted with saturated NaHCO<sub>3</sub> solution (32 mL). Most of the toluene was removed on the rotary evaporator before the mixture was extracted with diethyl ether  $(3 \times 30 \text{ mL})$ . The combined organic layers were dried over anhy-

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drous MgSO4, filtered, and concentrated in vacuo. Flash chromatography (petroleum ether/ethyl acetate 85:15) of the residue afforded pure carbamate 26 as a colorless oil (0.61 g, 45%).  $R_f$  = 0.45 (petroleum ether/ethyl acetate 70:30);  $[\alpha]_D^{25} = +51.9$   $(c=1.02, CH_2Cl_2)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 1.35$  (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.71 (dd, J=14.1, 4.8 Hz, 1H; CH<sub>2</sub>CO), 2.77-2.85 (m, 1H; CH<sub>2</sub>CO), 3.78 (s, 3H; OCH<sub>3</sub>), 4.19 (t, J= 6.8 Hz, 1H; CHCH<sub>2</sub>O), 4.38 (d,  $J=3.7$  Hz, 2H; CH<sub>2</sub>O), 5.07 (d,  $J=$ 4.5 Hz, 1 H; NHCH), 5.73 (d,  $J=6.3$  Hz, 1 H; NH), 6.85 (d,  $J=8.8$  Hz, 2H; aryl H), 7.21 (d,  $J=7.8$  Hz, 2H; aryl H), 7.28 (brs, 2H; aryl H), 7.38  $(t, J=7.3 \text{ Hz}, 2H; \text{ aryl H}), 7.57 \text{ (brs, 2H; aryl H)}, 7.74 \text{ ppm (d, } J=7.5,$ Hz, 2H; aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 27.9$  (C(CH<sub>3</sub>)<sub>3</sub>), 41.8 (CH<sub>2</sub>CO), 47.1 (CHCH<sub>2</sub>O), 51.3 (NHCH), 55.2 (OCH<sub>3</sub>), 66.6 (CH<sub>2</sub>O), 81.2 (C(CH3)3), 113.9, 119.9, 125.0, 126.9, 127.3, 127.6, 133.0, 141.2, 143.8, 155.5 (NHCO), 158.9, 170.1 ppm (CO<sub>2</sub>tBu); IR (film):  $\tilde{v} = 3332, 2973$ , 2256, 1720, 1527 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>29</sub>H<sub>31</sub>NO<sub>5</sub>: 474.22750; found: 474.22776  $[M+H]$ <sup>+</sup>.

tert-Butyl (3R)-3-amino-3-(4-methoxyphenyl)propanoate (27):  $Et<sub>2</sub>NH$  $(15 \text{ mL})$  at  $0^{\circ}$ C was slowly added to a solution of the Fmoc-protected amino acid ester 26 (0.271 g, 0.57 mmol) in dry THF (15 mL). Stirring was continued for 10 min at  $0^{\circ}$ C and then at room temperature for 12 h. The solution was concentrated in vacuo and the resulting oil purified by flash chromatography (methanol/dichloromethane/diethylamine 5:94.9:0.1). The pure amine 27 was obtained as colorless oil (0.086 g, 60%).  $R_f = 0.5$  (methanol/dichloromethane/diethylamine 5:94.9:0.1);  $[\alpha]_{\text{D}}^{25}$  = +12.3 (c = 1.00, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.42 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 1.81 (s, 2H; NH<sub>2</sub>), 2.54 (s, 1H; CH<sub>2</sub>), 2.56 (d, J=2.7 Hz, 1H; CH2), 3.79 (s, 3H; OCH3), 4.32 (dd, J=7.5, 6.0 Hz, 1H; CH), 6.86 (d,  $J=8.5$  Hz, 2H; aryl H), 7.27 ppm (d,  $J=8.5$  Hz, 2H; aryl H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 28.0 (C(CH<sub>3</sub>)<sub>3</sub>), 45.3 (CH<sub>2</sub>), 52.0 (CH), 55.1 (OCH<sub>3</sub>), 80.5 (C(CH<sub>3</sub>)<sub>3</sub>), 113.7, 127.3, 136.8, 158.7, 171.3 ppm (C= O); IR (film):  $\tilde{v} = 3378$ , 2973, 2842, 1724, 1157 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for  $C_{14}H_{21}NO_3$ : 252.15942; found: 252.15923  $[M+H]$ <sup>+</sup>.

### $tert$ -Butyl (3R)-3-({N-[(9H-fluoren-9-ylmethoxy)carbonyl]-p-tryptophyl}amino)-3-(4-methoxyphenyl)propanoate (28):

To a solution of amine  $27$  (0.080 g, 0.31 mmol) and N-Fmoc-D-tryptophan (0.135 g, 0.31 mmol) in anhydrous THF (2.2 mL) were added HOBt (0.043 g, 0.31 mmol) and DCC (0.095 g, 0.46 mmol) at  $-10\text{°C}$  followed by stirring of the mixture at 0°C overnight. The mixture was filtered through Celite and concentrated. The crude product was purified by flash chromatography (dichloromethane/methanol 98:2) to yield pure dipeptide 28 (0.192 g, 91%). M.p. 88.9–90.8 °C;  $R_f = 0.18$  (methanol/dichloromethane 2:98);  $[\alpha]_{\text{D}}^{25} = +7.64$  (c=1.03, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 1.22$  (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.50 (dd, J = 11.6, 5.3 Hz, 1H; Tyr CH<sub>2</sub>), 2.65 (dd,  $J=14.6$ , 4.3 Hz, 1H; Tyr CH<sub>2</sub>), 3.14 (dd,  $J=14.4$ , 7.8 Hz, 1H; Trp CH<sub>2</sub>), 3.31 (dd,  $J=12.1$ , 0.5 Hz, 1H; Trp CH<sub>2</sub>), 3.71, 3.75 (2 s, 3H; OCH<sub>3</sub>), 4.18 (br s, 1H; Fmoc CH), 4.31–4.44 (m, 2H; Fmoc CH<sub>2</sub>), 4.53 (d,  $J=3.0$  Hz, 1H; Trp CH), 5.22 (dd,  $J=14.4$ , 6.3 Hz, 1H; Tyr CH), 5.60 (d,  $J=5.0$  Hz, 1H; Trp NH), 6.57 (d,  $J=7.8$  Hz, 1H; Tyr NH), 6.71 (d,  $J=7.5$  Hz, 2H; Tyr aryl H), 6.78 (s, 1H; Ind NHCH), 6.89 (d,  $J=$ 8.5 Hz, 2H; Fmoc CH), 7.11 (t,  $J=7.4$  Hz, 1H; Trp aryl H), 7.19 (t,  $J=$ 7.5 Hz, 1H; Trp aryl H), 7.26–7.33 (m, 3H; Fmoc aryl H, Trp aryl H), 7.38 (t,  $J=7.4$  Hz, 2H; Fmoc aryl H), 7.53 (d,  $J=6.8$  Hz, 2H; Fmoc aryl H), 7.68 (d, J=6.5 Hz, 1H; Trp aryl H), 7.75 (d, J=7.5 Hz, 2H; Tyr aryl H), 8.03, 8.19 ppm (2s, 1H; Trp NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 27.6 (C(CH<sub>3</sub>)<sub>3</sub>), 28.6 (Trp CH<sub>2</sub>), 41.2 (Tyr CH<sub>2</sub>), 46.9 (Fmoc CH), 49.2 (Tyr CH), 55.0 (OCH<sub>3</sub>), 60.3 (Trp CH), 67.0 (Fmoc CH<sub>2</sub>), 81.0 (C(CH<sub>3</sub>)<sub>3</sub>), 109.8 (Trp aryl CH), 111.2 (Trp aryl CH), 113.6 (Tyr aryl CH), 118.5, (Trp aryl CH), 119.5 (Trp aryl CH), 119.8 (Fmoc aryl CH), 121.9 (Trp aryl CH), 123.2 (Trp NHCH), 125.0 (Fmoc aryl CH), 126.9 (Fmoc aryl CH), 127.1 (Trp aryl C), 127.3 (Fmoc aryl CH), 127.5 (Tyr aryl CH), 132.2 (Tyr aryl C), 136.1 (Trp aryl C), 141.1, (Fmoc aryl C), 143.6 (Fmoc aryl C), 155.9 (Fmoc CO), 158.6 (aryl C-OCH<sub>3</sub>), 169.9 (Tyr CO), 170.4 ppm (Trp CO); IR (film):  $\tilde{v} = 3313, 2935, 1716, 1515, 1245 \text{ cm}^{-1}$ ; HRMS (ESI):  $m/z$ : calcd for  $C_{40}H_{41}N_3O_6Na$ : 682.28876; found: 682.28960  $[M+Na]^+$ .

### N-[(9H-Fluoren-9-ylmethoxy)carbonyl]-l-alanyl-N-[(1R)-3-tert-butoxy-1- (4-methoxyphenyl)-3-oxopropyl]-D-tryptophanamide (29)

a)  $tert$ -Butyl  $(3R)$ -3-(4-methoxyphenyl)-3-( $p$ -tryptophylamino)propanoate: To a solution of the dipeptide 28 (0.134 g, 0.203 mmol) in dry THF (5 mL) was added Et<sub>2</sub>NH (15 mL) at  $0^{\circ}$ C and the mixture was stirred at  $0^{\circ}$ C for 15 min and then at room temperature for 1 h. Thereafter, the reaction mixture was concentrated in vacuo and the crude product purified by flash chromatography (methanol/dichloromethane/diethyl amine 2:97.9:0.1) to afford the free amine as a colorless solid (0.069 g, 78%). M.p. 138.7–140.5 °C;  $R_f$  = 0.2 (methanol/dichloromethane/diethylamine 2:97.9:0.1);  $[\alpha]_D^{23} = +43.7$  (c=0.21, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.32 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 1.52 (s, 2H; NH<sub>2</sub>), 2.59 (dd, J = 14.9, 6.5 Hz, 1 H; Tyr CH<sub>2</sub>), 2.76 (dd,  $J=14.9$ , 6.5 Hz, 1 H; Tyr CH<sub>2</sub>), 2.91 (dd,  $J=14.6$ , 9.0 Hz, 1H; Trp CH<sub>2</sub>), 3.38 (dd,  $J=14.4$ , 4.0 Hz, 1H; Trp CH<sub>2</sub>), 3.69 (dd,  $J=9.1$ ,  $J=4.3$  Hz, 1H; Trp CH), 3.76 (s, 3H; OCH<sub>3</sub>), 5.33 (dd,  $J=14.9, 6.5$  Hz, 1H; Tyr CH), 6.81 (d,  $J=8.5$  Hz, 2H; Tyr aryl H), 6.99  $(d, J=2.0 \text{ Hz}, 1 \text{ H}; \text{ Ind NHCH}), 7.09 (t, J=7.0 \text{ Hz}, 1 \text{ H}; \text{ Trp} \text{ aryl H}), 7.17$  $(t, J=6.8 \text{ Hz}, 1\text{ H};$  Trp aryl H), 7.17 (d,  $J=8.5 \text{ Hz}, 2\text{ H};$  Tyr aryl H), 7.33 (d,  $J=8.0$  Hz, 1H; Trp aryl H), 7.64 (d,  $J=7.8$  Hz, 1H; Trp aryl H), 8.02 (d,  $J=8.5$  Hz, 1H; Tyr NH CH), 8.50, 8.54 ppm (2s, 1H; Trp NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 27.8$  (C(CH<sub>3</sub>)<sub>3</sub>), 30.8 (Trp CH<sub>2</sub>), 41.7 (Tyr CH<sub>2</sub>), 48.9 (Tyr CH), 55.1 (OCH<sub>3</sub>), 55.5 (Trp CH), 81.0 (C(CH<sub>3</sub>)<sub>3</sub>), 111.2, 111.4, 113.8, 118.8, 119.4, 122.0, 123.1 (Trp NHCH), 127.3, 127.5, 132.9, 136.4 (Trp aryl C), 158.7, 170.0 (Tyr CO), 173.9 ppm (Trp CO); IR (film):  $\tilde{v} = 3309, 2923, 1720, 1658, 1249 \text{ cm}^{-1}$ ; HRMS (EI):  $m/z$ : calcd for  $C_{25}H_{31}N_3O_4$ : 438.23873; found: 438.23879  $[M+H]$ <sup>+</sup>.

b) Peptide coupling: To a solution of the amine described above (66 mg, 0.151 mmol) in dry THF (2 mL, 0.07m) were added Fmoc-l-Ala-OH (46 mg, 0.151 mmol), HOBt (20.3 mg, 0.151 mmol), and DCC (46 mg, 0.22 mmol) at  $0^{\circ}$ C and the mixture was stirred at  $0^{\circ}$ C for 12 h. Thereafter, the mixture was filtered and concentrated. Flash chromatography (methanol/dichloromethane 2:98) yielded the desired tripeptide 29 (0.110 g, 97%) as a slightly yellow solid. M.p. 116.2–123.5 °C;  $R_f = 0.4$ (methanol/dichloromethane 4:96);  $[a]_D^{25} = +10.17$  (c=1.03, CH<sub>2</sub>Cl<sub>2</sub>);<br><sup>1</sup>H NMP (400 MHz CDCl);  $\delta = 1.17, 1.21$  (m, 12H; C(CH), Ala CH) <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.17–1.21 (m, 12H; C(CH<sub>3</sub>)<sub>3</sub>, Ala CH<sub>3</sub>), 2.42 (dd,  $J=14.9$ , 6.3 Hz, 1H; Tyr CH<sub>2</sub>), 2.56 (dd,  $J=15.1$ , 6.5 Hz, 1H; Tyr CH<sub>2</sub>), 3.08 (dd,  $J=14.2$ , 6.9 Hz, 1H; Trp CH<sub>2</sub>), 3.20 (dd,  $J=14.6$ , 5.3 Hz, 1H; Trp CH2), 3.61 (s, 3H; OCH3), 4.03–4.11 (m, 2H; Ala CH, Fmoc CH), 4.15–4.29 (m, 2H; Fmoc CH<sub>2</sub>), 4.69 (brd,  $J=6.5$  Hz, 1H; Trp CH), 5.15 (dd,  $J=13.9$ , 6.5 Hz, 1H; Tyr CH), 5.54 (d,  $J=6.8$  Hz, 1H; Trp NH), 6.62 (d, J=8.5 Hz, 2H; Tyr aryl H), 6.78 (s, 1H; Ind NHCH), 6.82  $(d, J=7.5 \text{ Hz}, 1 \text{ H}; \text{ Ala NH}), 6.88$   $(d, J=8.3 \text{ Hz}, 2 \text{ H}; \text{ Tvr arvl H}), 6.94$   $(d,$  $J=8.0$  Hz, 1H; Tyr NH), 6.99 (t,  $J=7.5$  Hz, 1H; Trp aryl H), 7.05 (t,  $J=$ 7.3 Hz, 1H; Trp aryl H), 7.16–7.22 (m, 3H; Fmoc aryl H, Trp aryl H), 7.30 (t, J=7.3 Hz, 2H; Fmoc aryl H), 7.45 (d, J=6.1 Hz, 2H; Fmoc aryl H), 7.53 (d, J=7.3 Hz, 1H; Trp aryl H), 7.67 (d, J=7.3 Hz, 2H; Fmoc aryl H), 8.04, 8.11 ppm (2s, 1H; Ind NH); <sup>13</sup>C NMR (100 MHz, CDCl3):  $\delta$  = 18.3 (Ala CH<sub>3</sub>), 27.6 (Trp CH<sub>2</sub>), 27.7 ((CH<sub>3</sub>)<sub>3</sub>), 41.4 (Tyr CH<sub>2</sub>), 46.9 (Fmoc CH), 49.5 (Tyr CH), 50.8 (Ala CH), 53.6 (Trp CH), 55.1 (OCH3), 67.0 (Fmoc CH<sub>2</sub>), 81.1 (C(CH<sub>3</sub>)<sub>3</sub>), 110.1 (Trp aryl C), 111.1 (Trp aryl CH), 113.7 (Tyr aryl CH), 118.6, (Trp aryl CH), 119.6 (Trp aryl CH), 119.9 (Fmoc aryl CH), 122.1 (Trp aryl CH), 123.2 (Trp NHCH), 125.0 (Fmoc aryl CH), 127.0 (Fmoc aryl CH), 127.3 (Trp aryl C), 127.5 (Fmoc aryl CH), 127.7 (Tyr aryl CH), 132.5 (Tyr aryl C), 136.0 (Trp aryl C), 141.2 (Fmoc aryl C), 143.6 (Fmoc aryl C), 156.0 (Fmoc CO), 158.7 (aryl C), 170.0 (Tyr CO), 170.2 (Ala CO), 172.3 ppm (Trp CO); IR (film):  $\tilde{v} =$ 3401, 3289, 3062, 2931, 2117, 1697, 1646, 1245 cm<sup>-1</sup>; HRMS (ESI): m/z: calcd for  $C_{43}H_{46}N_4O_7Na$ : 753.32587; found: 753.32533  $[M+Na]^+$ .

N-[(2S)-3-(3-{(2R)-2-[(tert-Butoxycarbonyl)amino]propyl}phenyl)-2 methylpropanoyl]-L-alanyl-N-[(1R)-3-tert-butoxy-1-(4-methoxyphenyl)-3oxopropyl]-D-tryptophanamide (30)

a) L-Alanyl-N-[(1R)-3-tert-butoxy-1-(4-methoxyphenyl)-3-oxopropyl]-D**tryptophanamide:** Et<sub>2</sub>NH (6.5 mL) was added dropwise at  $0^{\circ}C$  to a solution of tripeptide 29 (162 mg, 0.222 mmol) in dry THF (6.5 mL, 0.034m). The mixture was then stirred for 15 min at  $0^{\circ}$ C and then at room temperature for 1 h. Thereafter, the mixture was concentrated and the oily residue purified by flash chromatography (methanol/dichloromethane/diethyl amine 2:97.9:0.1) yielding the free amine (0.096 g, 85%) as an oil.  $R_f$ = 0.11 (methanol/dichloromethane/diethylamine 2:97.9:0.1);  $[\alpha]_D^{25} = +23.9$  $(c=1.05, CH_2Cl_2)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 1.24–1.28 (m, 12 H;  $C(CH<sub>3</sub>)<sub>3</sub>$ , Ala CH<sub>3</sub>), 1.84 (s, 2H; NH<sub>2</sub>), 2.48 (dd, J = 15.1, 6.5 Hz, 1H; Tyr CH<sub>2</sub>), 2.65 (dd,  $J=15.1$ , 6.3 Hz, 1H; Tyr CH<sub>2</sub>), 3.17 (dd,  $J=14.6$ , 7.3 Hz,

1H; Trp CH<sub>2</sub>), 3.25 (dd,  $J=14.6$ , 6.3 Hz, 1H; Trp CH<sub>2</sub>), 3.40 (q,  $J=$ 6.9 Hz, 1 H; Ala CH), 3.72, 3.75 (2s, 3H; OCH<sub>3</sub>), 4.72 (q,  $J=7.2$  Hz, 1 H; Trp CH), 5.21 (dd, J=14.4, 6.5 Hz, 1H; Tyr CH), 6.74 (d, J=8.8 Hz, 2H; Tyr aryl H), 6.83 (s, 1H; Ind NHCH), 6.97 (d, J=8.5 Hz, 3H; Tyr aryl H, Tyr NH), 7.07 (t, J=7.2 Hz, 1H; Trp aryl H), 7.14 (t, J=7.2 Hz, 1H; Trp aryl H), 7.29 (d,  $J=8.08$  Hz, 1H; Trp aryl H), 7.64 (d,  $J=7.8$  Hz, 1H; Trp aryl H), 7.83 (d,  $J=8.08$  Hz, 1H; Trp NH), 8.29, 8.43 ppm (2s, 1H; Ind NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 21.3$  (Ala CH<sub>3</sub>), 27.7 C(CH<sub>3</sub>)<sub>3</sub>), 27.9 (Trp CH<sub>2</sub>), 41.2 (Tyr CH<sub>2</sub>), 49.3 (Tyr CH), 50.5 (Ala CH), 53.4 (Trp CH), 55.1 (OCH<sub>3</sub>), 81.1 (C(CH<sub>3</sub>)<sub>3</sub>), 110.2 (Trp aryl C), 111.2 (Trp aryl CH), 113.6 (Tyr aryl CH), 118.6 (Trp aryl CH), 119.3 (Trp aryl CH), 121.8 (Trp aryl CH), 123.2 (Trp NHCH), 127.4 (Tyr aryl CH, Trp aryl C), 132.4 (Tyr aryl C), 136.1 (Trp aryl C), 158.6 (aryl C-OCH3), 170.1 (Tyr CO), 170.3 (Ala CO), 175.9 ppm (Trp CO); IR (film):  $\tilde{v} = 3297, 3054,$ 2969, 2927, 1720, 1153 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>28</sub>H<sub>36</sub>N<sub>4</sub>O<sub>5</sub>: 509.27585; found: 509.27563 [M+H]<sup>+</sup>.

b) Peptide coupling: DCC (19.2 mg, 0.093 mmol) was added at  $-20^{\circ}$ C to a solution of amino acid 4 (19.5 mg, 0.061 mmol), the amine prepared above (31.0 mg, 0.061 mmol), and HOBt (8.2 mg, 0.061 mmol) in dry THF (0.8 mL). The mixture was stirred at  $-20$  °C for 30 min and then at room temperature for 8 h. Subsequently, the reaction mixture was filtered through Celite and the filtrate concentrated in vacuo. Purification of the residue by flash chromatography (methanol/dichloromethane 2:98) gave the tetrapeptide 30 as a white solid (0.046 g, 92%). M.p. 109.5–116.5 °C;  $R_{\rm f} = 0.10$  (methanol/dichloromethane 2:98);  $[\alpha]_{\rm D}^{25} = +33.1$  (c=1.00, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.96 (d, J = 6.3 Hz, 3H; CH<sub>3</sub>), 1.03 (d,  $J=5.8$  Hz, 3H; Ala CH<sub>3</sub>), 1.12 (d,  $J=7.0$  Hz, 3H; CH<sub>3</sub>), 1.33– 1.36 (2s, 18H; C(CH<sub>3</sub>)<sub>3</sub>), 2.39 (dd, J = 12.5, 9.2 Hz, 1H; CH<sub>2</sub>), 2.55–2.57 (m, 2H; CH<sub>2</sub>, Tyr CH<sub>2</sub>), 2.67-2.79 (m, 4H; CH<sub>2</sub>, CHCH<sub>3</sub>, Tyr CH<sub>2</sub>, Trp CH<sub>2</sub>), 2.85 (dd,  $J=13.3$ , 5.8 Hz, 1H; CH<sub>2</sub>), 3.06 (dd,  $J=14.4$ , 8.4 Hz, 1H; Trp CH<sub>2</sub>), 3.33-3.34 (2s, 3H; OCH<sub>3</sub>), 3.72-3.76 (m, 1H; CHCH<sub>3</sub>), 4.11-4.17 (m, 1H; Ala CH), 4.57–4.62 (m, 1H; Trp CH), 5.25 (dd, J=15.4, 7.2 Hz, 1H; Tyr CH), 6.74 (d, J=8.5 Hz, 2H; Tyr aryl H), 6.93–7.02 (m, 5H; Ind NHCH, Trp aryl H, xylyl H), 7.06 (t,  $J=7.4$  Hz, 1H; Trp aryl H), 7.08–7.12 (m, 3H; Tyr aryl H, xylyl H), 7.30 (d, J=8.0 Hz, IH, Trp aryl H), 7.51 (d, J=7.8 Hz, 1H; Trp aryl H), 7.82 (d, J=7.5 Hz, 1H; Trp NH), 8.07 (brs, 1H; Ala NH), 8.28 (d, J=8.3 Hz, 1H; Tyr NH), 10.27 ppm (s, 1H; Ind NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 17.4$  (CH<sub>3</sub>, Ala CH<sub>3</sub>), 20.6 (CH<sub>3</sub>), 28.4, 28.6, 28.9 (C(CH<sub>3</sub>)<sub>3</sub>, Trp CH<sub>2</sub>), 40.9 (CH<sub>2</sub>), 43.1 (CHCH<sub>2</sub>),43.2 (Tyr CH<sub>2</sub>), 43.9 (CH<sub>2</sub>), 51.3 (CHNBoc), 51.4 (Ala CH), 55.6 (Tyr CH), 55.6 (Trp CH), 55.7 (OCH<sub>3</sub>), 79.9, 82.1 (C(CH<sub>3</sub>)<sub>3</sub>), 111.1 (Trp aryl C), 112.4 (Trp aryl CH), 114.8 (Tyr aryl CH), 119.5, 119.9, 122.5 (Trp aryl CH), 124.7 (Ind NHCH), 128.0, 128.4 (xylyl CH), 128.8 (Trp aryl C), 129.2 (Tyr aryl CH), 129.3, 131.3 (xylyl CH), 134.4 (Tyr aryl C), 138.2 (Trp aryl C), 140.4, 140.1 (xylyl C), 157.8 (Boc CO), 160.5 (Tyr aryl C-OMe), 171.7, 172.9, (Ala CO, Tyr CO), 176.0 (Trp CO), 179.0 ppm (CO); IR (film):  $\tilde{v} = 3293$ , 2923, 1681, 1531 cm<sup>-1</sup>; HRMS (ESI):  $m/z$ : calcd for  $C_{46}H_{61}N_5O_8Na$ : 834.44124; found: 834.43921  $[M+Na]^+$ .

#### (3S,7R,10R,13S,16S)-10-(1H-Indol-3-ylmethyl)-7-(4-methoxyphenyl)- 3,13,16-trimethyl-4,8,11,14-tetraazabicyclo[16.3.1]docosa-1(22),18,20-

triene-5,9,12,15-tetrone (31): TFA (1.26 mL) was added dropwise at  $0^{\circ}$ C to a solution of tetrapeptide 30 (0.063 g, 0.0776 mmol) in dry  $CH_2Cl_2$ (1.26 mL) and the mixture stirred at room temperature. After HPLC-MS showed the complete deprotection (ca. 3 h), the mixture was concentrated on the rotary evaporator. The crude amino acid was used in the next step without further purification. It was dissolved in dry DMF (59 mL, 0.00098m) and treated with TBTU (56.0 mg, 0.17 mmol), HOBt (22.9 mg, 0.17 mmol), and  $iPr<sub>2</sub>NEt$  (0.040 mL, 0.23 mmol) at room temperature followed by stirring at this temperature for 14 h. The mixture was partitioned between ethyl acetate and water. The organic layer was washed with H<sub>2</sub>O, 5% aqueous KHSO<sub>4</sub> solution, H<sub>2</sub>O, half saturated NaHCO<sub>3</sub> solution, brine, dried over MgSO<sub>4</sub>, filtered, and concentrated in vacuo. Purification was achieved by flash chromatography (methanol/dichloromethane 2:98). The pure compound 31 was obtained as a colorless solid (0.046 g, 92%). M.p. 274.5–274.8 °C (decomp);  $R_f = 0.10$  (methanol/dichloromethane 2:98);  $\left[\alpha\right]_D^{25} = 5.5$  (c=0.45, DMSO); <sup>1</sup>H NMR (600 MHz, [D<sub>6</sub>]DMSO):  $\delta = 0.87$  (d, J = 6.1 Hz, 3H; 5-CH<sub>3</sub>), 0.91 (d, J = 7.0 Hz, 3H; 14-CH<sub>3</sub>), 1.04 (d,  $J=7.0$  Hz, 3H; 11-CH<sub>3</sub>), 2.22 (dd,  $J=12.7$ , 11.0 Hz, 1H; 6-H), 2.35–2.49 (m, 4H; 2-H, 10-H, 2-H, 11-H), 2.78–2.85 (m, 2H; 6-H,

20-H), 2.95 (dd,  $J=13.2$ , 10.5 Hz, 1H; 10-H), 3.01 (dd,  $J=14.0$ , 5.3 Hz, 1H; 20-H), 3.73 (s, 1H; OCH3), 3.74 (m, 1H; 5-H), 4.46–4.51 (m, 1H; 14-H), 4.62–4.66 (m, 1H; 17-H), 5.13–5.17 (m, 1H; 1-H), 6.82–6.86 (m, 2H; 2''-H, 2'-H), 6.88 (d, J=7.0 Hz, 1H; 6''-H), 6.91–6.95 (m, 2H; 4''-H, 7'''-H), 6.96–6.99 (m, 1H; 2'''-H), 7.03 (dd, J=7.5, 7.5 Hz, 1H; 8'''-H), 7.06–7.12 (m, 2H; 5'''-H, 3'-H), 7.31 (d, J=7.5 Hz, 1H; 9'''-H), 7.54 (d,  $J=7.9$  Hz, 2H; 6"'-H, 4-NH), 7.82 (d,  $J=7.9$  Hz, 1H; 13-NH), 8.21 (d, J=8.8 Hz, 1H; 16-NH), 8.50 (d, J=8.8 Hz, 1H; 19-NH), 10.77 ppm (s, 1H; 1'''-NH); <sup>13</sup>C NMR (150 MHz,  $[D_6]$ DMSO):  $\delta$  = 18.7 (CH<sub>3</sub>-5), 18.8  $(CH_3$ -14), 19.4 (CH<sub>3</sub>-11), 28.4 (C-20), 38.7 (C-10), 41.1 (C-11), 42.3 (C-6), 42.7 (C-2), 46.0 (C-5), 47.0 (C-14), 49.3 (C-1), 52.6 (C-17), 54.8 (OCH3), 110.9 (C-9'''), 117.8 (C-7'''), 118.0 (C-6'''), 120.5 (C-8'''), 123.3 ppm (C- $2^{\prime\prime\prime}$ ); IR (film):  $\tilde{v} = 3286, 3062, 2965, 2927, 1643, 1542 \text{ cm}^{-1}$ ; HRMS (ESI):  $m/z$ : calcd for C<sub>37</sub>H<sub>43</sub>N<sub>5</sub>O<sub>5</sub>: 660.31564; found: 660.31495 [M+Na]<sup>+</sup>.

#### Methyl N-(tert-butoxycarbonyl)-L-alanyl-N-methyl-D-tryptophanate (33)

a) N-Me-D-Trp-OMe: A solution of N-Boc-protected amino acid ester<sup>[3d]</sup> 32 (2.042 g, 6.15 mmol) in dry  $CH_2Cl_2$  (50 mL) was treated dropwise with TFA  $(5.00 \text{ mL}, 0.0615 \text{ mol})$  at  $0^{\circ}\text{C}$  and the mixture stirred at room temperature for 1 h. The mixture was concentrated and the residue dissolved in ethyl acetate, then washed with water. The aqueous layer was basified with 1<sub>N</sub> NaOH to pH 8–9 keeping the temperature at  $0^{\circ}$ C. Both layers were then extracted with ethyl acetate. The organic layer was dried over anhydrous MgSO4, filtered, and concentrated to yield the free amine (0.975 g, 68%). It was used in the next step without further purification. <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 2.37 (s, 3H; NCH<sub>3</sub>), 3.12 (dd, J = 14.4, 7.0 Hz, 1 H; CH<sub>2</sub>), 3.21 (dd,  $J=14.4$ , 5.8 Hz, 1 H; CH<sub>2</sub>), 3.59 (t,  $J=6.0$  Hz, 1H; CH), 3.66 (s, 3H; OCH3), 6.96 (d, J=2.02 Hz, 1H; Ind NHCH), 7.11  $(t, J=7.8 \text{ Hz}, 1 \text{ H}; \text{ aryl H}), 7.17 (t, J=7.8 \text{ Hz}, 1 \text{ H}; \text{ aryl H}), 7.29 (d, J=$ 7.8 Hz, 1H; aryl H), 7.60 (d, J=7.8 Hz, 1H; aryl H), 8.70 ppm (s, 1H; Ind NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 28.9 (CH<sub>2</sub>), 34.6 (NCH<sub>3</sub>), 51.6 (OCH<sub>3</sub>), 63.6 (CH<sub>2</sub>CH), 110.4 (aryl C), 111.1, 118.4, 119.2, 121.8, 123.0, (aryl CH), 127.2, 136.1, (aryl C), 174.9 ppm (CO<sub>2</sub>); IR (film):  $\tilde{v} = 3405-$ 2803 (broad), 1731, 1446, 1203 cm<sup>-1</sup>.

b) Peptide coupling: Boc-l-Ala-OH (0.305 g, 1.61 mmol), HOBt (0.218 g, 1.61 mmol), and DCC (0.50 g, 2.42 mmol) were added at  $0^{\circ}$ C to a solution of the foregoing amine (0.375 g, 1.61 mmol) in dry THF (14 mL). The reaction mixture was stirred at  $0^{\circ}$ C overnight and then filtered through Celite, and concentrated. Purification of the residue by flash chromatography (dichloromethane/methanol 98:2) gave the dipeptide 33 as a colorless solid (0.384 g, 59%). M.p. 73.5–77.4 °C;  $R_f = 0.16$  (dichloromethane/methanol 98:2);  $[\alpha]_D^{26} = +54.33$   $(c=1.05, CH_2Cl_2)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 0.88$  (d, J = 6.8 Hz, 3H; Ala CH<sub>3</sub>), 1.41 (s, 9H; C- $(CH<sub>3</sub>)<sub>3</sub>$ , 2.81 (s, 3H; NCH<sub>3</sub>), 3.28 (dd, J=15.4, 11.1 Hz, 1H; CH<sub>2</sub>), 3.45 (dd,  $J=15.4$ , 4.8 Hz, 1H; CH<sub>2</sub>), 3.73 (s, 3H; OCH<sub>3</sub>), 4.44–4.52 (m, 1H; Ala CH), 5.26 (dd, J=10.9, 4.9 Hz, 1H; Trp CH), 5.50 (d, J=7.8 Hz, 1H; NH), 6.98 (d,  $J=1.7$  Hz, 1H; Ind NHCH), 7.10 (t,  $J=7.4$  Hz, 1H; aryl H), 7.16 (t,  $J=7.0$  Hz, 1H; aryl H), 7.31 (d,  $J=7.8$  Hz, 1H; aryl H), 7.56 (d,  $J=7.8$  Hz, 1H; aryl H), 8.34 ppm (s, 1H; IndNH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 18.1$  (Ala CH<sub>3</sub>), 24.4 (CH<sub>2</sub>), 28.3 (C(CH<sub>3</sub>)<sub>3</sub>), 32.8  $(NCH_3)$ , 46.5 (Ala CH), 52.3 (OCH<sub>3</sub>), 58.1 (Trp CH), 79.5 (C(CH<sub>3</sub>)<sub>3</sub>), 110.6 (aryl C), 111.2, 118.2, 119.4, 122.0 (aryl CH), 122.4 (Ind NHCH), 127.0, 136.0 (aryl C), 155.1 (NHCO<sub>2</sub>), 171.1 (Ala CO), 173.5 ppm (Trp CO); IR (film):  $\tilde{v} = 3926, 2973, 1704, 1646, 1488 \text{ cm}^{-1}$ ; HRMS (EI):  $m/z$ : calcd for  $C_{21}H_{29}N_3O_5Na$ : 426.19994; found: 426.2000  $[M+Na]^+$ .

N-(tert-Butoxycarbonyl)-L-alanyl-N-methyl-D-tryptophan (34): NaOH solution  $(0.4 \text{ m} \text{ in H}_2\text{O}, 0.024 \text{ g}, 0.60 \text{ mmol})$  was added to a solution of the dipeptide 33 (0.201 g, 0.48 mmol) in THF (3 mL). The reaction mixture was stirred for 2 h at room temperature, and then diluted with saturated NaHCO<sub>3</sub> solution (7.5 mL). The mixture was then extracted with  $(3 \times$ 4 mL) of diethyl ether. The aqueous layer was acidified to pH 1–2 with  $0.5$ <sub>N</sub> HCl at  $0$ <sup>o</sup>C. The acidified layer was extracted with ethyl acetate  $(3 \times 15 \text{ mL})$ . The combined organic layers were dried over anhydrous MgSO4, filtered, and concentrated in vacuo. The crude acid was purified by flash chromatography (ethyl acetate/petroleum ether/acetic acid 1:1:0.1) yielding 34 as a pale colorless solid (0.141 g, 74%). M.p. 106.7– 107.4 °C;  $R_f = 0.12$  (ethyl acetate/petroleum ether/acetic acid 1:1:0.1);  $[\alpha]_{\text{D}}^{25}$  = +44.84 (c = 1.17, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 0.86  $(d, J=6.8 \text{ Hz}, 3\text{ H}; \text{ Ala } \text{CH}_3)$ , 1.42 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.80 (s, 3H; NCH<sub>3</sub>),

3.32 (dd,  $J=15.1$ , 11.3 Hz, 1H; CH<sub>2</sub>), 3.48 (dd,  $J=15.6$ , 4.8 Hz, 1H; CH2), 4.49–4.56 (m, 1H; Ala CH), 5.23 (dd, J=10.4, 4.4 Hz, 1H; Trp CH), 5.71 (d,  $J=8.0$  Hz, 1H; NH), 6.99 (brs, 1H; Ind NHCH), 7.10 (t, J=7.3 Hz, 1H; aryl H), 7.16 (t, J=7.3 Hz, 1H; aryl H), 7.34 (d, J= 7.8 Hz, 1H; aryl H), 7.56 (d, J=7.8 Hz, 1H; aryl H), 8.55, 8.61 (s, 1H; Ind NH), 10.62 ppm (s, 1H; CO<sub>2</sub>H); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 17.6 (Ala CH<sub>3</sub>), 24.2 (CH<sub>2</sub>), 28.3 (C(CH<sub>3</sub>)<sub>3</sub>), 33.3 (NCH<sub>3</sub>), 46.6 (Ala CH), 58.9 (Trp CH), 79.8 (C(CH<sub>3</sub>)<sub>3</sub>), 110.4 (aryl C), 111.3, 118.1, 119.3, 121.9 (aryl CH), 122.7 (Ind NHCH), 127.0, 136.0 (aryl C), 155.4 (Boc CO), 173.8 (Ala CO), 174.2 ppm (Trp CO); IR (film):  $\tilde{v} = 3012, 1762, 1700$ , 1463, 1373, 1245 cm<sup>-1</sup>; HRMS (ESI):  $m/z$ : calcd for C<sub>20</sub>H<sub>27</sub>N<sub>3</sub>O<sub>5</sub>Na: 412.18429; found: 412.18462 [M+Na]<sup>+</sup>.

N-(tert-Butoxycarbonyl)-l-alanyl-N-[(1R)-3-tert-butoxy-1-(4-methoxyphenyl)-3-oxopropyl]-N-methyl-D-tryptophanamide (35): HOBt (0.135 g, 0.33 mmol) and DCC (0.101 g, 0.49 mmol) at  $-20$ °C were added to a solution of acid 34 (0.128 g, 0.33 mmol) and amine 27 (0.082 g, 0.33 mmol) in dry THF (1.5 mL). The mixture was stirred at  $0^{\circ}$ C overnight, filtered, and the filtrate concentrated. The crude tripeptide was purified by flash chromatography (ethyl acetate/petroleum ether 50:50) providing 35 as a colorless solid (0.196 g, 96%). M.p. 109.2–112.6 °C;  $R_f$  = 0.25 (ethyl acetate/petroleum ether 1:1);  $[\alpha]_D^{25} = +21.85$  (c=1.02, CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 0.86$  (d, J = 6.8 Hz, 3H; Ala CH<sub>3</sub>), 1.33 (s, 9H; C- $(CH<sub>3</sub>)<sub>3</sub>$ ), 1.39 (s, 9H; C(CH<sub>3</sub>)<sub>3</sub>), 2.65 (dd, J = 15.4, 6.0 Hz, 1H; Tyr CH<sub>2</sub>), 2.78 (dd,  $J=15.6$ ,  $8.0$  Hz,  $1$  H; Tyr CH<sub>2</sub>),  $2.91$  (s,  $3$  H; NCH<sub>3</sub>),  $3.18$  (dd,  $J=15.6$ , 10.6 Hz, 1 H; Trp CH<sub>2</sub>), 3.42 (dd,  $J=15.1$ , 5.5 Hz, 1 H; Trp CH<sub>2</sub>), 3.74 (s, 3H; OCH3), 4.33–4.39 (m, 1H; Ala CH), 5.29–5.34 (m, 2H; Tyr CH, NH), 5.60 (dd,  $J=10.3$ , 5.5 Hz, 1H; Trp), 6.77 (d,  $J=8.5$  Hz, 2H; Tyr aryl H), 6.92 (br s, 1H; NHCH), 7.04 (d, J=8.0 Hz, 1H; NH), 7.08 (t,  $J=7.4$  Hz, 1H; Trp aryl H), 7.15–7.16 (m, 3H; Tyr aryl H, Trp aryl H), 7.29 (d, J=8.1 Hz, 1H; Trp aryl H), 7.56 (d, J=7.5 Hz, 1H; Trp aryl H), 8.52 ppm (s, 1H; Ind NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 16.9 (Ala CH<sub>3</sub>), 23.16 (Trp CH<sub>2</sub>), 27.8, 28.2 (C(CH<sub>3</sub>)<sub>3</sub>), 30.6 (NCH<sub>3</sub>), 41.5 (Tyr CH2), 46.5 (Ala CH), 49.5 (Tyr CH), 55.1 (OCH3), 56.5 (Trp CH), 79.6, 80.8 (C(CH3)3), 110.7 (Trp aryl C-8), 111.0 (Trp aryl CH), 113.7 (Tyr aryl CH), 118.4, 119.1, 121.7 (Trp aryl CH), 122.0 (Ind NHCH), 127.3 (Trp aryl C), 127.5 (Tyr aryl CH), 132.9 (Tyr aryl C), 136.0, (Trp aryl C), 155.4 (Boc CO), 158.7 (Tyr aryl C-OCH3), 169.1 (Tyr CO), 169.8 (Ala CO), 174.2 ppm (Trp CO); IR (film):  $\tilde{v} = 3332, 2977, 1666, 1515, 1160 \text{ cm}^{-1}$ ; HRMS (ESI):  $m/z$ : calcd for C<sub>34</sub>H<sub>46</sub>N<sub>4</sub>O<sub>7</sub>Na: 645.32587; found: 645.32523  $[M+Na]^+$ .

### N-[(2S)-3-(3-{(2R)-2-[(tert-Butoxycarbonyl)amino]propyl}phenyl)-2 methylpropanoyl]-L-alanyl-N-[(1R)-3-tert-butoxy-1-(4-methoxyphenyl)-3oxopropyl]-N-methyl-D-tryptophanamide (36)

a) L-Alanyl-N-[(1R)-3-tert-butoxy-1-(4-methoxyphenyl)-3-oxopropyl]-Nmethyl-D-tryptophanamide: A solution of tripeptide 35 (0.120 g, 0.19 mmol) and  $2.6$ -lutidine  $(0.134 \text{ mL}$ , 1.16 mmol) in dry CH<sub>2</sub>Cl<sub>2</sub> (2.8 mL) was treated with TBDMSOTf (0.17 mL, 0.76 mmol) at  $0^{\circ}$ C. After 1 h of stirring at 0°C, HPLC-MS analysis showed that the tert-butyl residue was replaced by the tert-butyldimethylsilyl group. The reaction was then quenched by adding  $H_2O$  (8 mL). The mixture was extracted with ethyl acetate  $(3 \times 15 \text{ mL})$ . The combined organic layers were dried over anhydrous MgSO4, filtered, and concentrated. The crude product was purified by flash chromatography using a stepwise gradient (dichloromethane/methanol, 100:0; 98:2; 97:3; 90:10). The free amine was obtained as yellow oil (0.057 g, 56%).  $R_f$ =0.12 (dichloromethane/methanol/diethyl amine 95.9:4:0.1);  $\left[\alpha\right]_D^{26} = +16.65$  ( $c = 0.65$ , CH<sub>2</sub>Cl<sub>2</sub>); <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 0.68$  (d,  $J = 6.8$  Hz, 3H; Ala CH<sub>3</sub>), 1.30–1.36 (m, 11H; C(CH<sub>3</sub>)<sub>3</sub>, NH<sub>2</sub>), 2.67 (dd, J = 15.4, 5.5 Hz, 1H; Tyr CH<sub>2</sub>), 2.78 (dd,  $J=15.1, 7.1$  Hz, 1H; Tyr CH<sub>2</sub>), 2.83 (s, 3H; NCH<sub>3</sub>), 3.19 (dd,  $J=15.6$ , 11.1 Hz, 1 H; Trp CH<sub>2</sub>), 3.41 (dd,  $J=15.5$ , 5.0 Hz, 1 H; Trp CH<sub>2</sub>), 3.72 (s, 3H; OCH3), 3.88–3.93 (m, 1H; Ala CH), 5.31–5.36 (m, 1H; Tyr CH), 5.55 (dd, J=10.9, 5.4 Hz, 1H; Trp CH), 6.76 (d, J=8.6 Hz, 2H; Tyr aryl H), 6.90 (br s, 1H; Ind NHCH), 7.05 (d,  $J=7.4$  Hz, 1H; Tyr aryl H), 7.12 (t, J=7.2 Hz, 1H; Trp aryl H), 7.20 (d, J=8.5 Hz, 2H; Tyr aryl H), 7.27 (d,  $J=7.6$  Hz, 1H; Trp aryl H), 7.45 (d,  $J=8.1$  Hz, 1H; NH), 8.52 ppm (s, 1H; Ind NH); <sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta = 17.0$  (Ala CH<sub>3</sub>), 23.4 (Trp CH<sub>2</sub>), 27.8 (C(CH<sub>3</sub>)<sub>3</sub>), 30.5 (NCH<sub>3</sub>), 41.1 (Tyr CH<sub>2</sub>), 47.3 (Ala CH), 49.8 (Tyr CH), 55.2 (OCH<sub>3</sub>), 57.3 (Trp CH), 81.2 (C(CH<sub>3</sub>)<sub>3</sub>), 110.0 (Trp

aryl C), 111.3 (Trp aryl CH), 113.8 (Tyr aryl CH), 118.1, 119.2, 121.8 (Trp aryl CH), 122.5 (Ind NHCH), 127.3 (Trp aryl C), 127.8 (Tyr aryl CH), 133.0 (Tyr aryl C), 135.9 (Trp aryl C), 158.8 (Tyr aryl C-OCH<sub>3</sub>), 169.0 (Tyr CO), 170.7 (Ala CO), 174.2 ppm (Trp CO); IR (film):  $\tilde{v} = 3324$ , 2965, 1666, 1515 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>29</sub>H<sub>38</sub>N<sub>4</sub>O<sub>5</sub>: 523.29150; found: 523.29154 [M+H]<sup>+</sup>.

b) Peptide coupling: A solution of acid  $4$  (0.029 g, 0.056 mmol), the amine prepared above (0.018 g, 0.056 mmol), and HOBt (0.0075 g, 0.056 mmol) in dry THF  $(0.7 \text{ mL})$  was treated with DCC  $(0.017 \text{ g})$ , 0.084 mmol) at  $-20$  °C. The reaction mixture was first stirred at the same temperature for 30 min and then at room temperature for 18 h. The precipitate of urea was then filtered off and the filtrate concentrated on the rotary evaporator. Purification of the residue by flash chromatography (dichloromethane/methanol 96:4) gave tetrapeptide 36 as a colorless solid (0.028 g, 60%). M.p. 78.8-85.2°C;  $R_f$  = 0.12 (dichloromethane/methanol 94:6);  $[\alpha]_D^{25} = +30.45$   $(c=0.57, CH_2Cl_2)$ ; <sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta = 0.84$  (d,  $J = 6.8$  Hz, 3H; Ala CH<sub>3</sub>), 1.00 (d,  $J = 6.8$  Hz, 3H; CH<sub>3</sub>), 1.04 (d, J = 6.3 Hz, 3H; CH<sub>3</sub>), 1.35–1.37 (2s, 18H; C(CH<sub>3</sub>)<sub>3</sub>), 2.39 (dd,  $J=13.3$ , 8.6 Hz, 1H; CH<sub>2</sub>), 2.54–2.61 (m, 2H; CH<sub>2</sub>, Tyr CH<sub>2</sub>), 2.63– 2.72 (m, 1H; CHCH<sub>3</sub>), 2.75 (dd,  $J=15.5$ , 6.1 Hz, 1H; Tyr CH<sub>2</sub>), 2.90–3.03  $(m, 5H; NCH_3, CH_2), 3.12$  (dd,  $J=15.5, 11.2$  Hz, 1H; Trp CH<sub>2</sub>), 3.47 (dd,  $J=15.4$ , 5.1 Hz, 1H; Trp CH<sub>2</sub>), 3.66 (s, 3H; OCH<sub>3</sub>), 3.70–3.76 (m, 1H; CHCH3), 4.42–4.49 (m, 1H; Ala CH), 5.30–5.35 (m, 1H; Tyr CH), 5.58 (dd,  $J=11.1$ , 5.0 Hz, 1H; Trp CH), 6.75 (d,  $J=8.6$  Hz, 2H; Tyr aryl H), 6.95 (s, 1H; Ind NHCH), 6.97 (s, 1H; xylyl H), 6.99–7.01 (m, 3H; Trp aryl H, xylyl H), 7.05 (t,  $J=7.5$  Hz, 1H; Trp aryl H), 7.14 (t,  $J=7.5$  Hz, 1H; xylyl H), 7.19 (d,  $J=8.6$  Hz, 2H; Tyr aryl H), 7.29 (d,  $J=8.1$  Hz, 1H; Trp aryl H), 7.53 (d, J=7.8 Hz, 1H; Trp aryl H), 8.06 (d, J=8.3 Hz, 2H; Tyr NH, Ala NH), 10.26 ppm (s, 1H; Ind NH); 13C NMR (100 MHz, CDCl<sub>3</sub>):  $\delta$  = 16.1 (Ala CH<sub>3</sub>), 17.8 (CH<sub>3</sub>), 20.9 (CH<sub>3</sub>), 24.8 (Trp CH<sub>2</sub>), 28.4, 28.9 (C(CH<sub>3</sub>)<sub>3</sub>), 32.0 (NCH<sub>3</sub>), 40.9 (CH<sub>2</sub>), 43.0 (CH<sub>2</sub>), 43.1 (Tyr CH<sub>2</sub>), 44.0 (CH<sub>2</sub>), 47.3(Ala CH), 51.3 (CHNBoc), 51.4 (Tyr CH), 55.8 (OCH<sub>3</sub>), 58.7 (Trp CH), 79.8, 82.0 (C(CH<sub>3</sub>)<sub>3</sub>), 111.3 (Trp aryl C), 112.3 (Trp aryl CH), 114.9 (Tyr aryl CH), 119.3, 119.8, 122.5 (Trp aryl CH), 124.0 (Trp NHCH), 128.0, 128.4 (xylyl CH), 128.7 (Trp aryl C), 129.1 (Tyr aryl CH), 129.4, 131.3 (xylyl CH), 134.7 (Tyr aryl C), 138.1 (Trp aryl C), 140.5, 141.3 (xylyl C), 157.8 (Boc CO), 160.5 (Tyr aryl C-OMe), 171.7, 171.8, (Ala CO, Tyr CO), 176.2 (Trp CO), 179.0 ppm (CO); IR (film):  $\tilde{v} = 3309$ , 2973, 1650, 1519, 1160 cm<sup>-1</sup>; HRMS (EI):  $m/z$ : calcd for C<sub>47</sub>H<sub>63</sub>N<sub>5</sub>O<sub>8</sub>: 848.45689; found: 848.45733 [M+Na]<sup>+</sup>.

### (3S,7R,10R,13S,16S)-10-(1H-Indol-3-ylmethyl)-7-(4-methoxyphenyl)- 3,11,13,16-tetramethyl-4,8,11,14-tetraazabicyclo[16.3.1]docosa-

1(22),18,20-triene-5,9,12,15-tetrone (37): TFA (0.5 mL, 6.13 mmol) was added at  $0^{\circ}$ C to a solution of tetrapeptide 36 (0.025 g, 0.0303 mmol) in dry  $CH_2Cl_2$  (0.5 mL). The reaction mixture was stirred at room temperature for 1.5 h, before the solvents were evaporated on the rotary evaporator. The crude amine salt was used in the next step without further purification.

To a solution of the crude amino acid salt (0.0303 mmol) in dry DMF (30 mL, 0.001m) were added TBTU (0.029 g, 0.090 mmol), HOBt (0.0122 g, 0.090 mmol), and  $iPr<sub>2</sub>NEt$  (0.020 mL, 0.121 mmol) followed by stirring of the mixture at room temperature for 14 h. The mixture was partitioned between ethyl acetate and water. The combined organic layers were washed with water,  $5\%$  aqueous  $KHSO<sub>4</sub>$  solution, water, half saturated NaHCO<sub>3</sub> solution, water, brine, dried over anhydrous MgSO<sub>4</sub>, filtered, and concentrated. Purification of the residue by flash chromatography (dichloromethane/methanol 96:4) furnished the desired macrocycle 37 as a white powder (8.9 mg, 45%). M.p. 169.7–177 °C (decomp);  $R_f$  = 0.41 (dichloromethane/methanol 94:6);  $[a]_D^{25} = 6.5$  (c=0.45, DMSO);<br><sup>1</sup>H NMP (600 MHz ID IDMSO);  $\delta = 0.60$  (d,  $I = 7.0$  Hz 3 H; Ala CH) <sup>1</sup>H NMR (600 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 0.60 (d, J = 7.0 Hz, 3H; Ala CH<sub>3</sub>), 0.94 (d,  $J=6.1$  Hz, 3H; C-5 CH<sub>3</sub>), 1.04 (d,  $J=7.0$  Hz, 3H; C-11 CH<sub>3</sub>), 2.33 (dd, J=12.7, 9.2 Hz, 1H; 6-H), 2.37 (dd, J=14.0, 2.6 Hz, 1H; 10-H), 2.41–2.47 (m, 2H; 2-H), 2.49 (m, 1H; 11-H), 2.76 (dd,  $J=13.1$ , 4.4 Hz, 1H; 6-H), 2.92 (s, 3H; NMe), 2.96 (dd, J=14.0, 10.5 Hz, 1H; 10-H), 3.04 (dd, J=14.9, 11.4 Hz, 1H; 20-H), 3.13–3.17 (m, 1H; 20-H), 3.72 (s, 3H; OCH<sub>3</sub>), 3.72-3.78 (m, 1H; 5-H), 4.56 (ddd,  $J=13.8$ , 6.8, 6.6 Hz, 1H; 14-H), 5.15–5.19 (m, 1H; 1-H), 5.49 (dd, J=11.0, 4.8 Hz, 1H; 17-H), 6.85– 6.90 (m, 3H; 2''-H, 2'-H, 6''-H), 6.91–6.96 (m, 2H; 4''-H, 7'''-H), 6.99–7.00

 $(m, 1H; 2'''-H), 7.02$  (dd,  $J=7.5, 7.5$  Hz,  $1H; 8'''-H$ ), 7.08 (dd,  $J=7.5$ , 7.5 Hz, 1H; 5''-H), 7.18 (d, J=8.8 Hz, 1H; 3'-H), 7.29 (d, J=7.9 Hz, 1H; 9'''-H), 7.56 (d, J=7.9 Hz, 1H; 6'''-H), 7.63 (d, J=7.0 Hz, 1H; 4-NH), 7.86 (d,  $J=7.9$  Hz, 1H; 13-NH), 8.43 (d,  $J=7.9$  Hz, 1H; 19-NH), 10.77 ppm (br s, 1H; 1‴-NH); <sup>13</sup>C NMR (150 MHz, [D<sub>6</sub>]DMSO):  $\delta$  = 16.5 (Ala CH<sub>3</sub>), 18.9 (5 CH<sub>3</sub>), 19.6 (11 CH<sub>3</sub>), 24.3 (C-20), 30.0 (NCH<sub>3</sub>), 38.4 (C-10), 41.1 (C-11), 41.7 (C-2), 42.0 (C-6), 44.6 (C-14), 46.2 (C-5), 48.7 (C-1), 54.8 (OCH3), 55.1 (C-17), 110.9 (C-9'''), 117.8 (C-7'''), 117.9 (C-6'''), 120.5 (C-8'''), 122.7 ppm (C-8'''); HRMS (ESI): m/z: calcd for  $C_{38}H_{45}N_5O_5$ : 674.33129; found: 674.33154  $[M+Na]^+$ .

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- [1] For some recent examples, see: a) S. Hanessian, M. Angiolini, Chem. Eur. J. 2002, 8, 111 – 117; b) M. P. Glenn, M. J. Kelso, J. D. A. Tyndall, D. P. Fairlie, J. Am. Chem. Soc. 2003, 125, 640 – 641; c) J. Becerril, M. Bolte, M. I. Burguete, F. Galindo, E. Gracía-Espana, S. V. Luis, J. F. Miravet, J. Am. Chem. Soc. 2003, 125, 6677 – 6686; d) F. Liu, H.-Y. Zha, Z.-J. Yao, J. Org. Chem. 2003, 68, 6679 – 6684; e) R. C. Reid, G. Abbenante, S. M. Taylor, D. P. Fairlie, J. Org. Chem. 2003, 68, 4464 – 4471; f) Y. Hitotsuyanagi, S. Motegi, T. Hasuda, K. Takeya, Org. Lett. 2004, 6, 1111 – 1114; g) A. G. M. Barrett, A. J. Hennessy, R. Le Vézouët, P. A. Procopiou, P. W. Seale, S. Stefaniak, R. J. Upton, A. J. P. White, D. J. Williams, J. Org. Chem. 2004, 69, 1028 – 1037; h) P. Cristau, M.-T. Martin, M.-E. T. H. Dau, J.-P. Vors, J. Zhu, Org. Lett. 2004, 6, 3183 – 3186; i) H. B. Lee, M. C. Zaccaro, M. Pattarawarapan, S. Roy, H. U. Saragovi, K. Burgess, J. Org. Chem. 2004, 69, 701-713; j) D. S. Reddy, D. Vander Velde, J. Aubé, J. Org. Chem. 2004, 69, 1716-1719.
- [2] a) T. M. Zabriskie, J. A. Klocke, C. M. Ireland, A. H. Marcus, T. F. Molinski, D. J. Faulkner, C. Xu, J. C. Clardy, J. Am. Chem. Soc. 1986, 108, 3123 – 3124; b) P. Crews, L. V. Manes, M. Boehler, Tetrahedron Lett. 1986, 27, 2797 – 2800.
- [3] For total syntheses of 1, see: a) P. A. Grieco, Y. S. Hon, A. Perez-Medrano, J. Am. Chem. Soc. 1988, 110, 1630-1631; b) K. S. Chu, G. R. Negrete, J. P. Konopelski, J. Org. Chem. 1991, 56, 5196 – 5201; c) A. V. Rama Rao, M. K. Gurjar, B. R. Nallaganchu, A. Bhandari, Tetrahedron Lett. 1993, 34, 7085-7088; d) Y. Hirai, K. Yokota, T. Momose, Heterocycles 1994, 39, 603 – 612; e) P. Ashworth, B. Broadbelt, P. Jankowski, P. Kocienski, A. Pimm, R. Bell, Synthesis 1995, 199 – 206.
- [4] a) M. R. Bubb, I. Spector, B. B. Beyer, K. M. Fosen, J. Biol. Chem. 2000, 275, 5163-5170; b) A. Holzinger, Methods Mol. Biol. 2001, 161, 109 – 120.
- [5] W. R. Chan, W. F. Tinto, P. S. Manchand, L. J. Todaro, J. Org. Chem. 1987, 52, 3091 – 3093.
- [6] For total syntheses of geodiamolide, see: a) Y. Hirai, K. Yokota, T. Yamazaki, T. Momose, Heterocycles 1990, 30, 1101 – 1119; b) P. A.

Grieco, A. Perez-Medrano, Tetrahedron Lett. 1988, 29, 4225 – 4228; c) T. Shioiri, T. Imaeda, Y. Hamada, Heterocycles 1997, 46, 421 – 442; d) J. D. White, J. C. Amedio, J. Org. Chem. 1989, 54, 736 – 738.

- [7] R. N. Sonnenschein, J. J. Farias, K. Tenney, S. L. Mooberry, E. Lobkovsky, J. Clardy, P. Crews, Org. Lett. 2004, 6, 779 – 782.
- [8] For some key papers on conformational design, see: a) R. W. Hoffmann, M. Stahl, U. Schopfer, G. Frenking, Chem. Eur. J. 1998, 4, 559 – 566; b) R. W. Hoffmann, D. Stenkamp, T. Trieselmann, R. Göttlich, Eur. J. Org. Chem. 1999, 2915 – 2927; c) R. W. Hoffmann, Angew. Chem. 2000, 112, 2134 – 2150; Angew. Chem. Int. Ed. 2000, 39, 2054-2070; d) R. W. Hoffmann, R. Göttlich, U. Schopfer, Eur. J. Org. Chem. 2001, 1865 – 1871.
- [9] For syntheses of the acid 3, see: a) U. Schmidt, W. Siegel, K. Mundinger, Tetrahedron Lett. 1988, 29, 1269 – 1270; b) S.-K. Kang, D.-H. Lee, Synlett 1991, 175 – 176; c) A. V. Rama Rao, M. K. Gurjar, B. R. Nallaganchu, A. Bhandari, Tetrahedron Lett. 1993, 34, 7081 – 7084; d) S. Wattanasereekul, M. E. Maier, Adv. Synth. Catal. 2004, 346, 855 – 861.
- [10] For a review about carbohydrate-based amino acids, see: S. A. W. Gruner, E. Locardi, E. Lohof, H. Kessler, Chem. Rev. 2002, 102,  $491 - 514$ .
- [11] S. Terracciano, I. Bruno, G. Bifulco, J. E. Copper, C. D. Smith, L. G. Paloma, R. Riccio, J. Nat. Prod. 2004, 67, 1325 – 1331.
- [12] M. P. Trova, Y. Wang, Tetrahedron 1993, 49, 4147-4158.
- [13] For recent reviews about group selective reactions, see: a) L.-M. Zhu, M. C. Tedford, Tetrahedron 1990, 46, 6587 – 6611; b) M. E. Maier, Nachr. Chem. Tech. Lab. 1993, 41, 314 – 330; c) C. S. Poss, S. L. Schreiber, Acc. Chem. Res. 1994, 27, 9-17; d) S. R. Magnuson, Tetrahedron 1995, 51, 2167-2213; e) K. Mori, Synlett 1995, 1097-1109; f) M. C. Willis, J. Chem. Soc. Perkin Trans. 1 1999, 1765 – 1784; g) R. W. Hoffmann, Angew. Chem. 2003, 115, 1128-1142; Angew. Chem. Int. Ed. 2003, 42, 1096 – 1109; h) E. G. Urdiales, I. Alfonso, V. Gotor, Chem. Rev. 2005, 105, 313 – 354.
- [14] J. R. Gage, D. A. Evans, Org. Synth. 1989, 68, 77 82.
- [15] H. Iding, B. Wirz, R.-M. Rodríguez Sarmiento, Tetrahedron: Asymmetry 2003, 14, 1541-1545.
- [16] a) J. R. McDermott, N. L. Benoiton, Can. J. Chem. 1973, 51, 1915– 1919; b) S. T. Cheung, N. L. Benoiton, Can. J. Chem. 1977, 55, 906 – 910.
- [17] For a review, see: L. Aurelio, R. T. C. Brownlee, A. B. Hughes, Chem. Rev. 2004, 104, 5823 – 5846.
- [18] M. Prashad, H.-Y. Kim, D. Har, O. Repic, T. J. Blacklock, Tetrahedron Lett. 1998, 39, 9369 – 9372.
- [19] D. A. Evans, L. D. Wu, J. J. M. Wiener, J. S. Johnson, D. H. B. Ripin, J. S. Tedrow, J. Org. Chem. 1999, 64, 6411-6417.
- [20] M. Sakaitani, Y. Ohfune, Tetrahedron Lett. 1985, 26, 5543 5546.
- [21] W. Inman, P. Crews, J. Am. Chem. Soc. 1989, 111, 2822-2829.
- [22] The assays were performed in the group of Dr. Florenz Sasse, GBF-German Research Centre for Biotechnology, Chemical Biology, Mascheroder Weg 1, 38124 Braunschweig, Germany.
- [23] a) J. R. McDermott, N. L. Benoiton, Can. J. Chem. 1973, 51, 1915-1919; b) S. T. Cheung, N. L. Benoiton, Can. J. Chem. 1977, 55, 906-910.
- [24] For a review, see: L. Aurelio, R. T. C. Brownlee, A. B. Hughes, Chem. Rev. 2004, 104, 5823 – 5846.

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